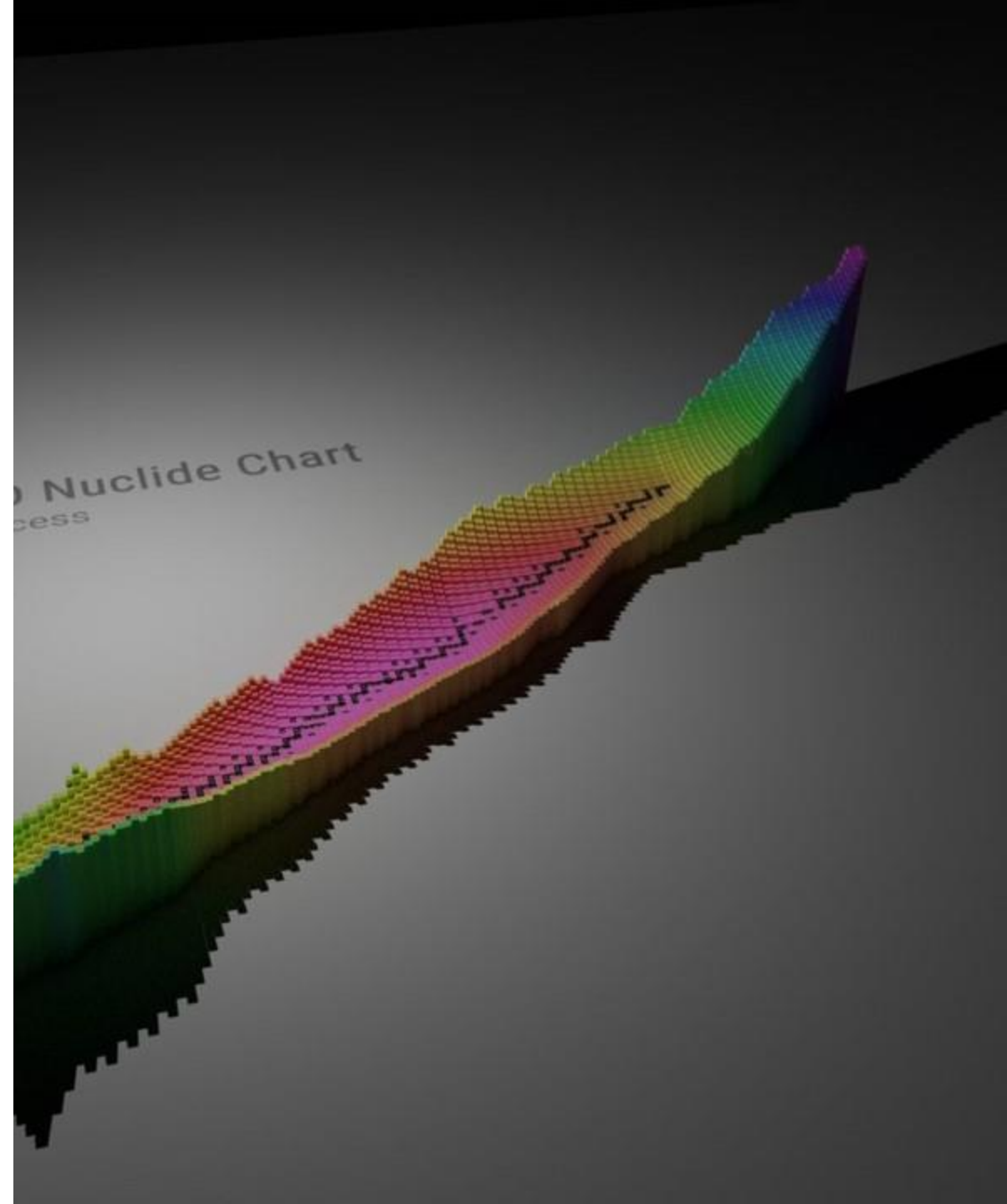


Secular disequilibrium in materials: implications on background estimates for rare event detectors

November 30th, 2022

Kirby Hobbs

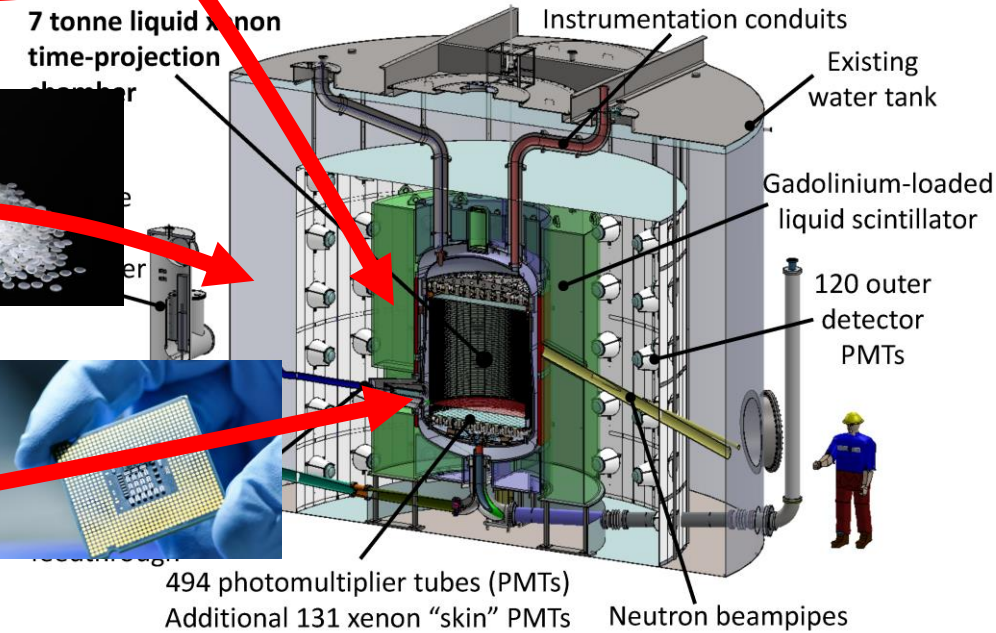
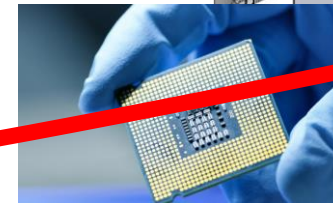
Isaac Arnquist
M. Laura Di Vacri
Tyler Schlieder



^{238}U and ^{232}Th and their progeny



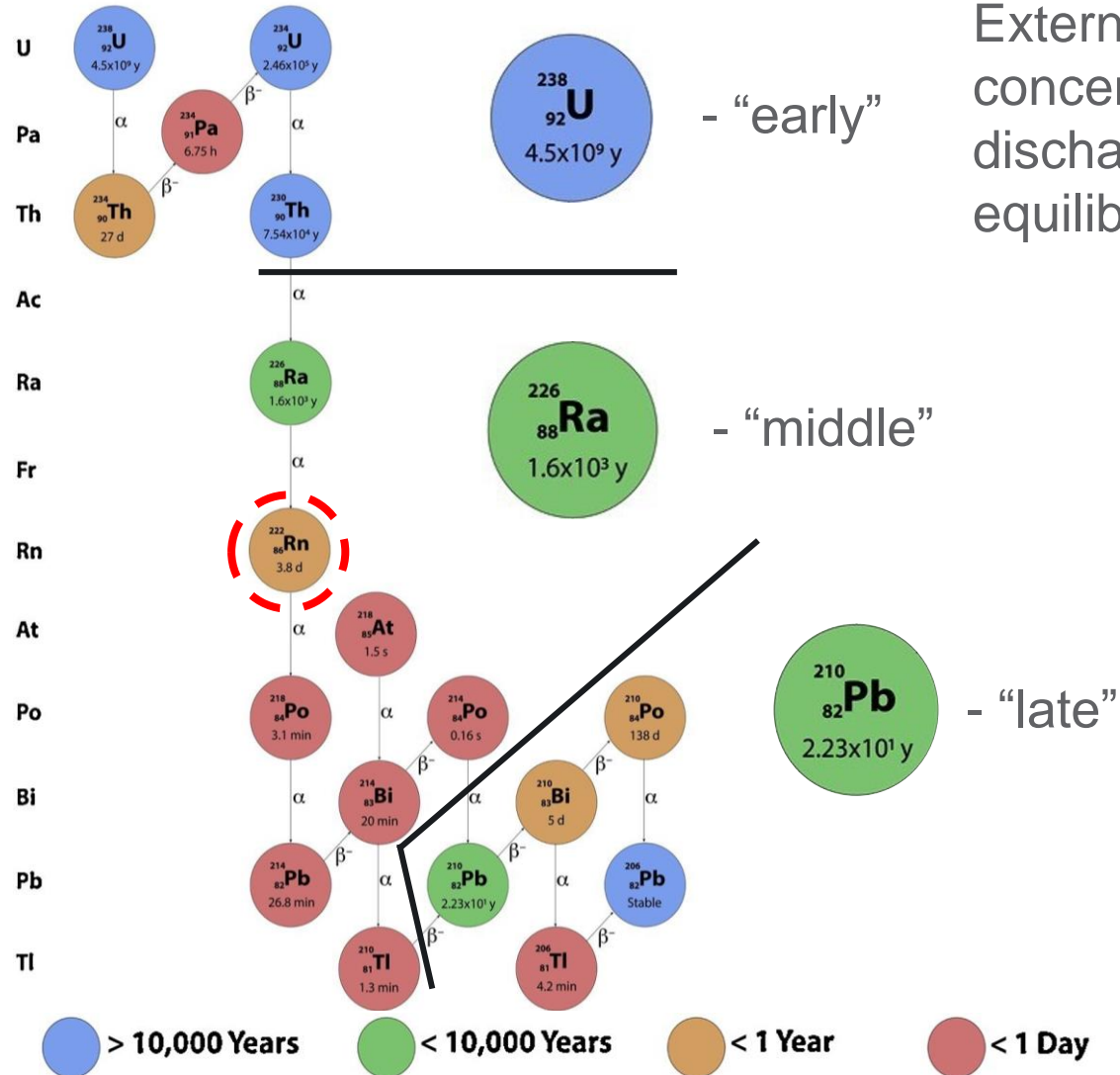
The LZ Detector



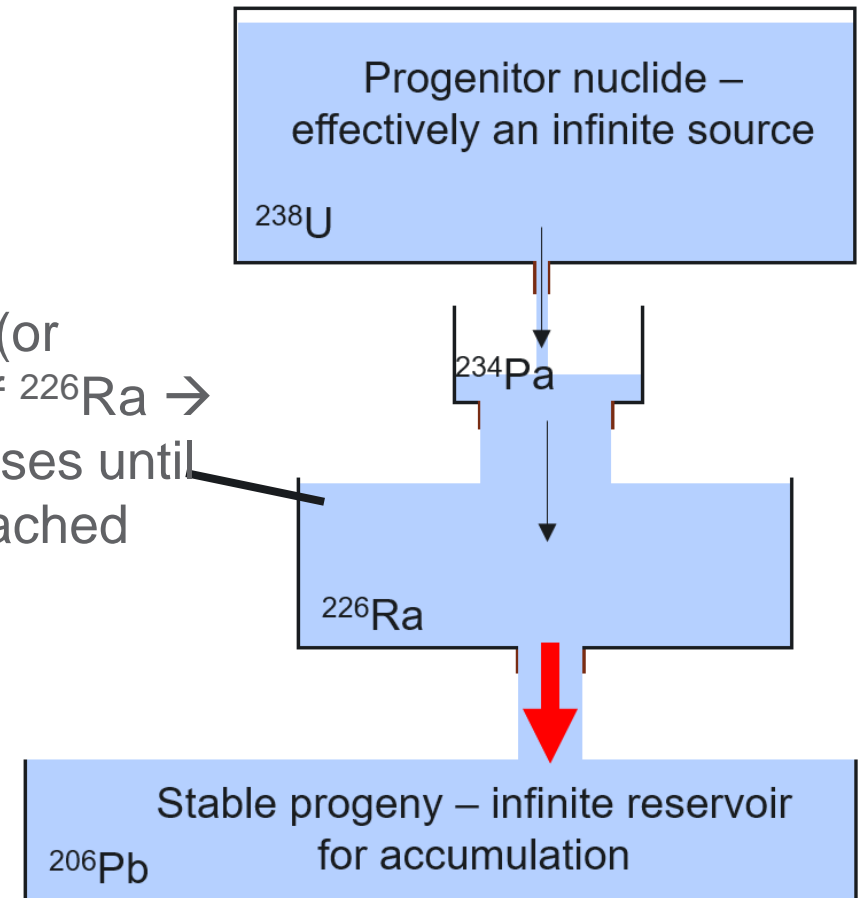
Sec equilibrium, *sans* radon

Sec equilibrium????

Secular Equilibrium

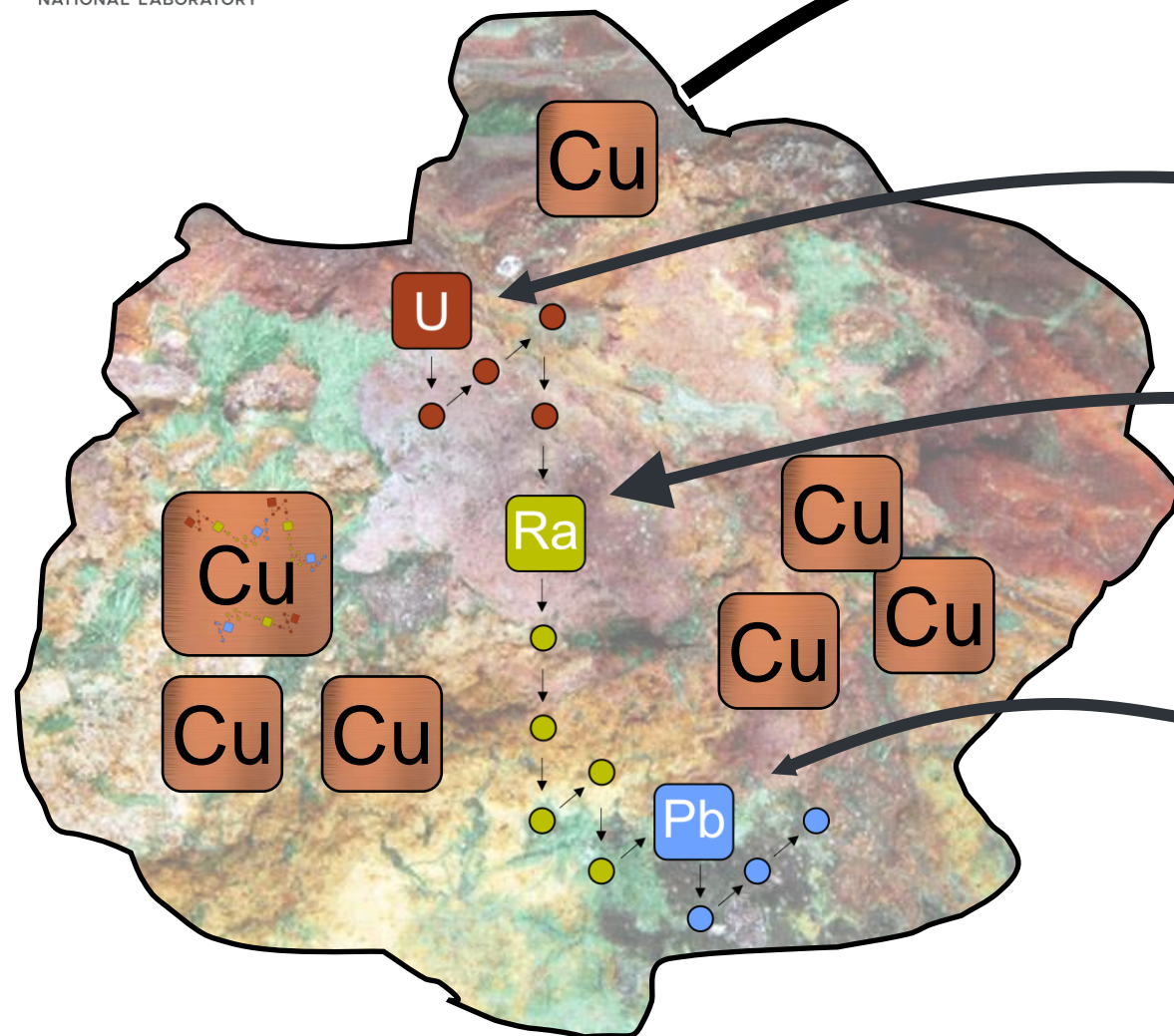


External source (or concentration) of $^{226}\text{Ra} \rightarrow$ discharge increases until equilibrium is reached

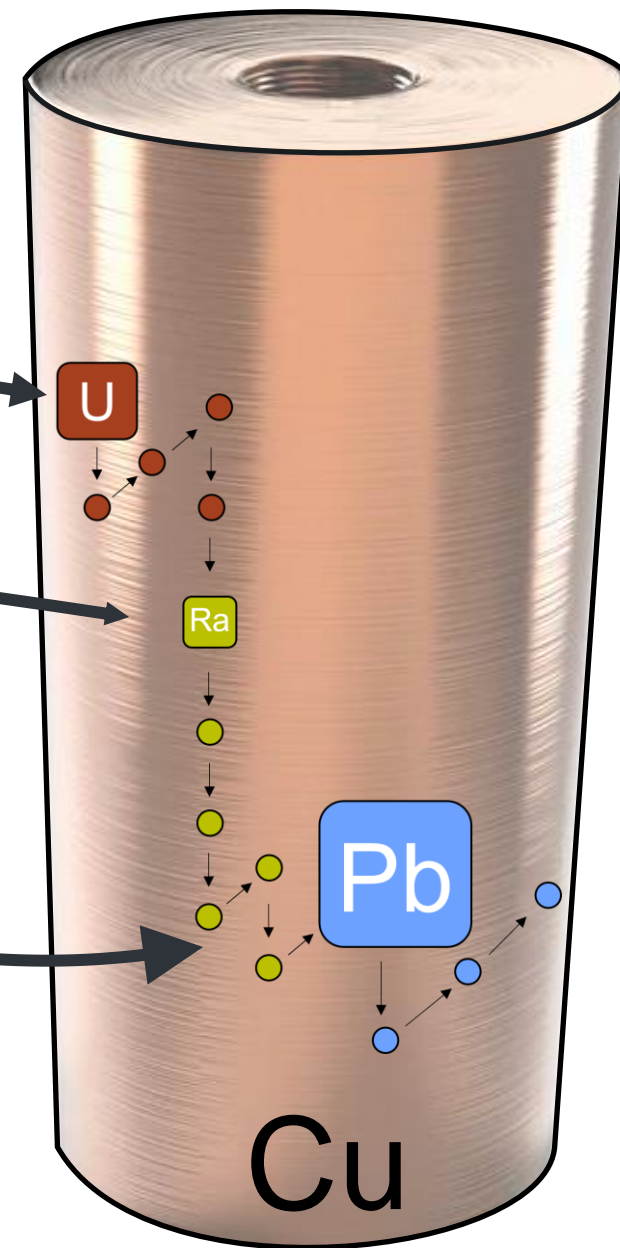


- Secular equilibrium – discharge from each bucket is equal
- Short half-life = large hole
- Long half-life = small hole

Smelting/refining



Raw Material (ore)



Refined product (Cu anode)

Motivation

Raw material/ore

Grinding
crushing
flotation

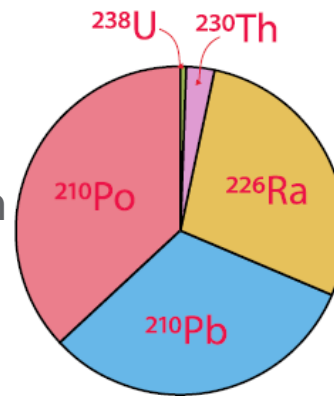
Cu-rich
concentrate



% RN Bq/g

Sulfuric acid leach

Leached Cu-rich
residue



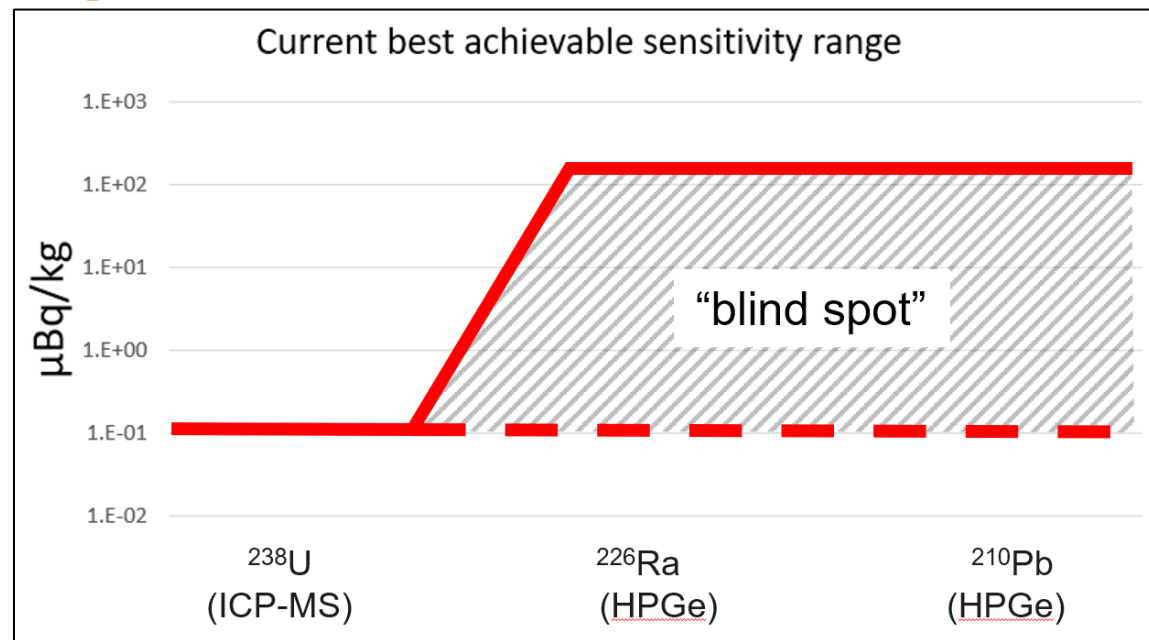
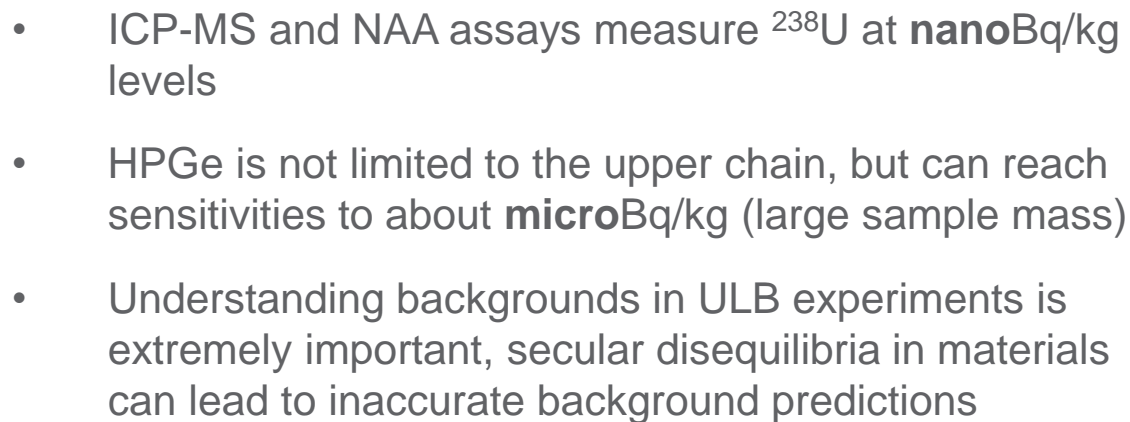
% RN Bq/g

Cu metal

Smelting
refining

Secular disequilibrium has been observed in starting material for copper concentrates from the Olympic Dam mine (fourth largest copper mine)

- Would **further** processing “push” the ^{238}U decay chain back into equilibrium?
- Evolution over different processes (e.g., smelting, refinement, electrochemical deposition, etc.)

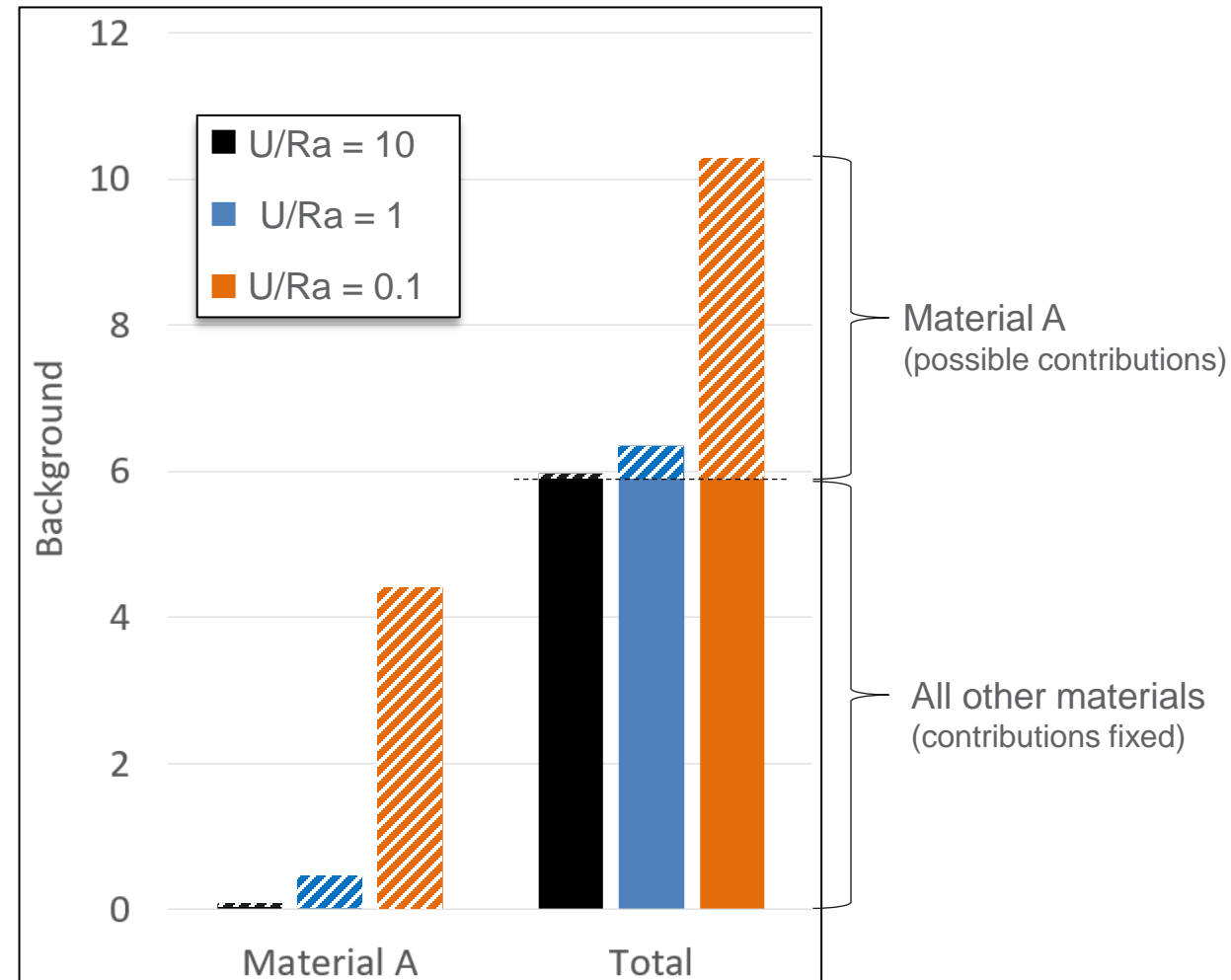


- ICP-MS and NAA assays measure ^{238}U at **nanoBq/kg** levels
- HPGe is not limited to the upper chain, but can reach sensitivities to about **microBq/kg** (large sample mass)
- Understanding backgrounds in ULB experiments is extremely important, secular disequilibria in materials can lead to inaccurate background predictions

Potential Impact on ULB Detectors

- Example of how secular equilibrium could affect background of a detector
 - Material A normally contributes 10% towards background when in equilibrium ($U = Ra$)
 - If $U < Ra$ by 10x, Material A contributes upwards of 50% towards background
 - Small part + small fractionation = not a big deal
 - Small part + large fractionation = problem

Detector X:



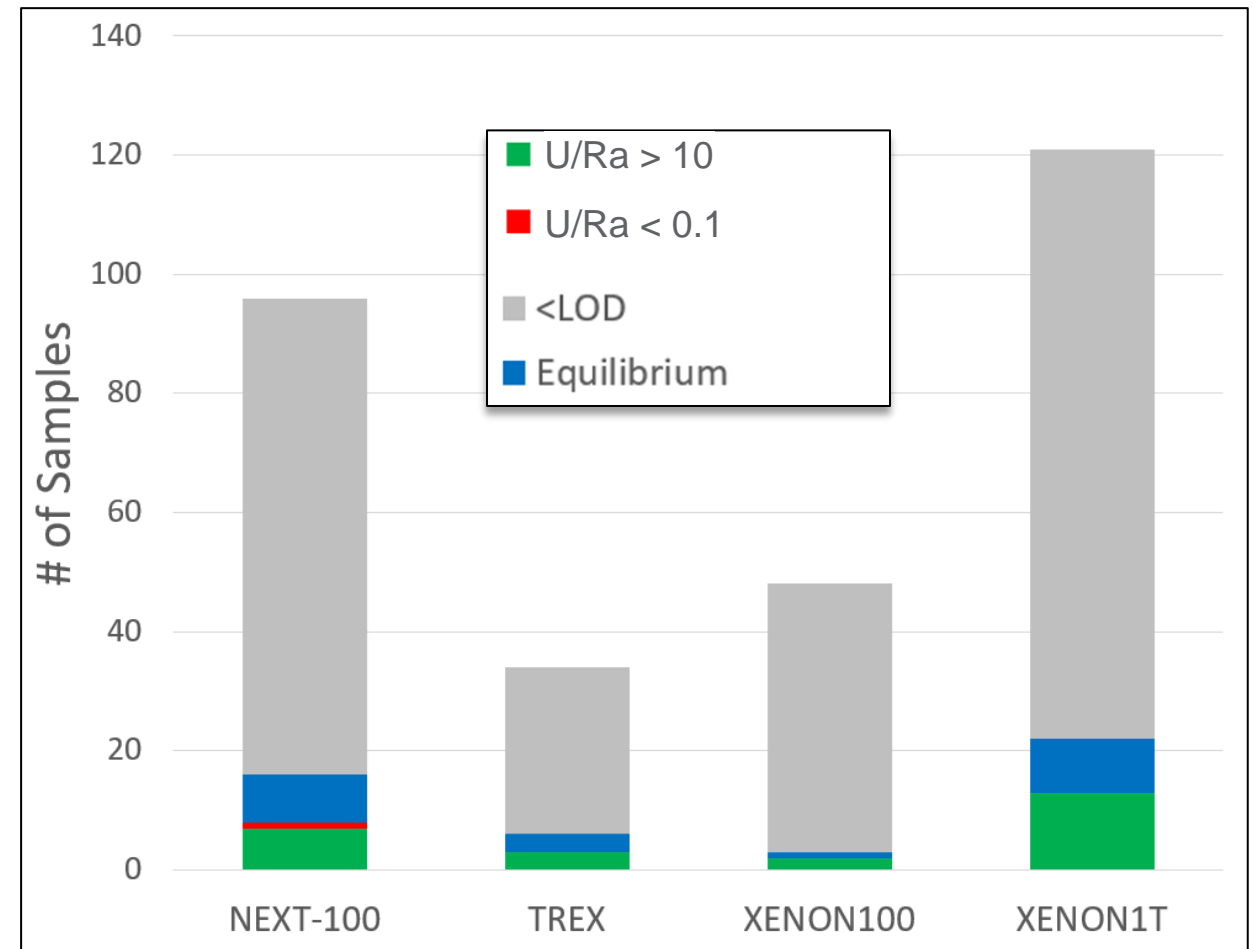
Secular Equilibrium in ULB Detectors

U/Ra fractionation:

- <LOD – secular equilibrium could not be determined due to instrumental detection limits or high measurement variability

Next generation detectors:

- increase in mass
- more stringent radiopurity requirements

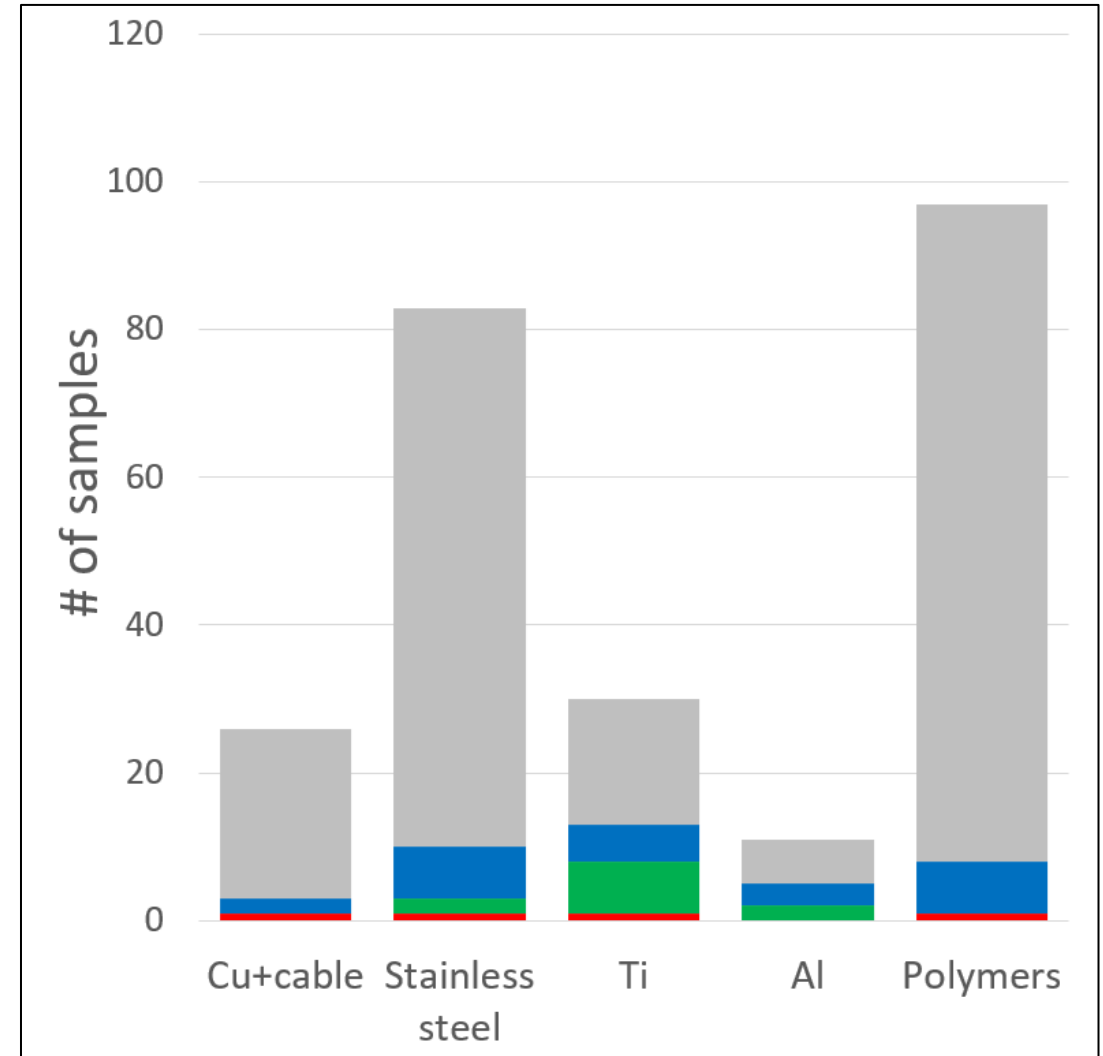
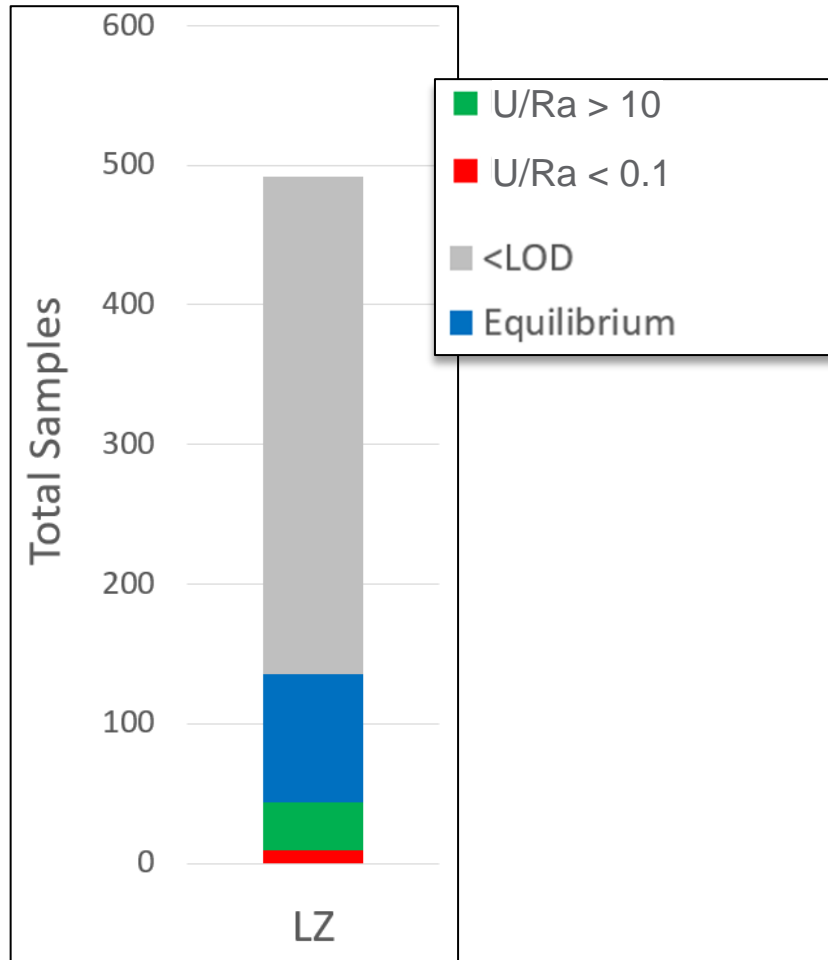


* Garnered from:

V. Alvarez et al. 2013
E. Aprile et al. 2011
E. Aprile et al. 2017
J. Castel et al. 2019

LUX-Zeplin (LZ) Assay Data – U/Ra

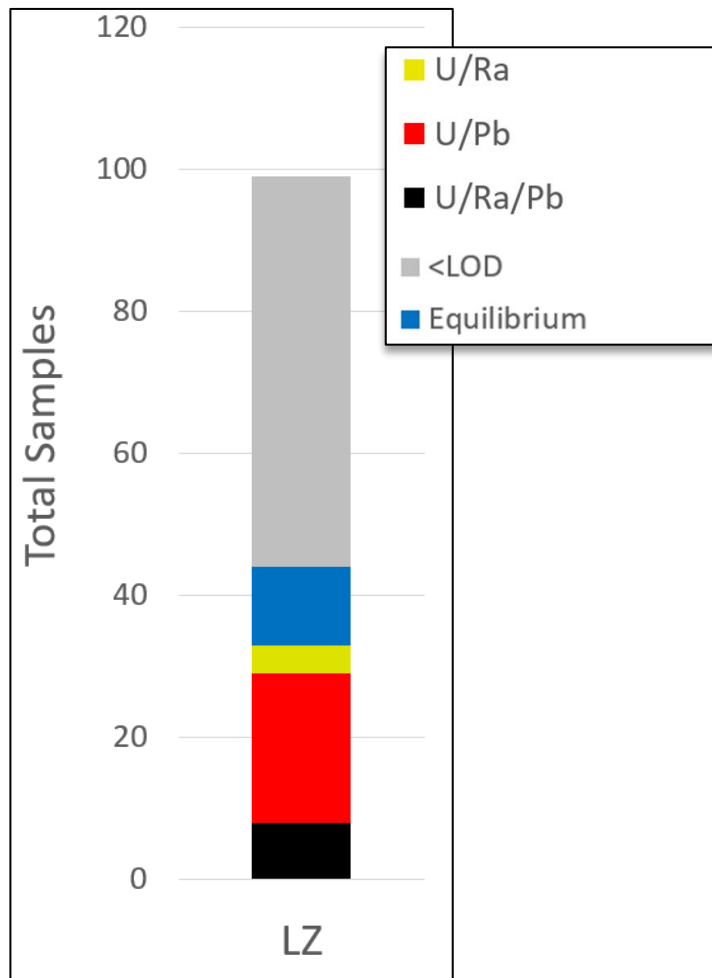
U/Ra fractionation:



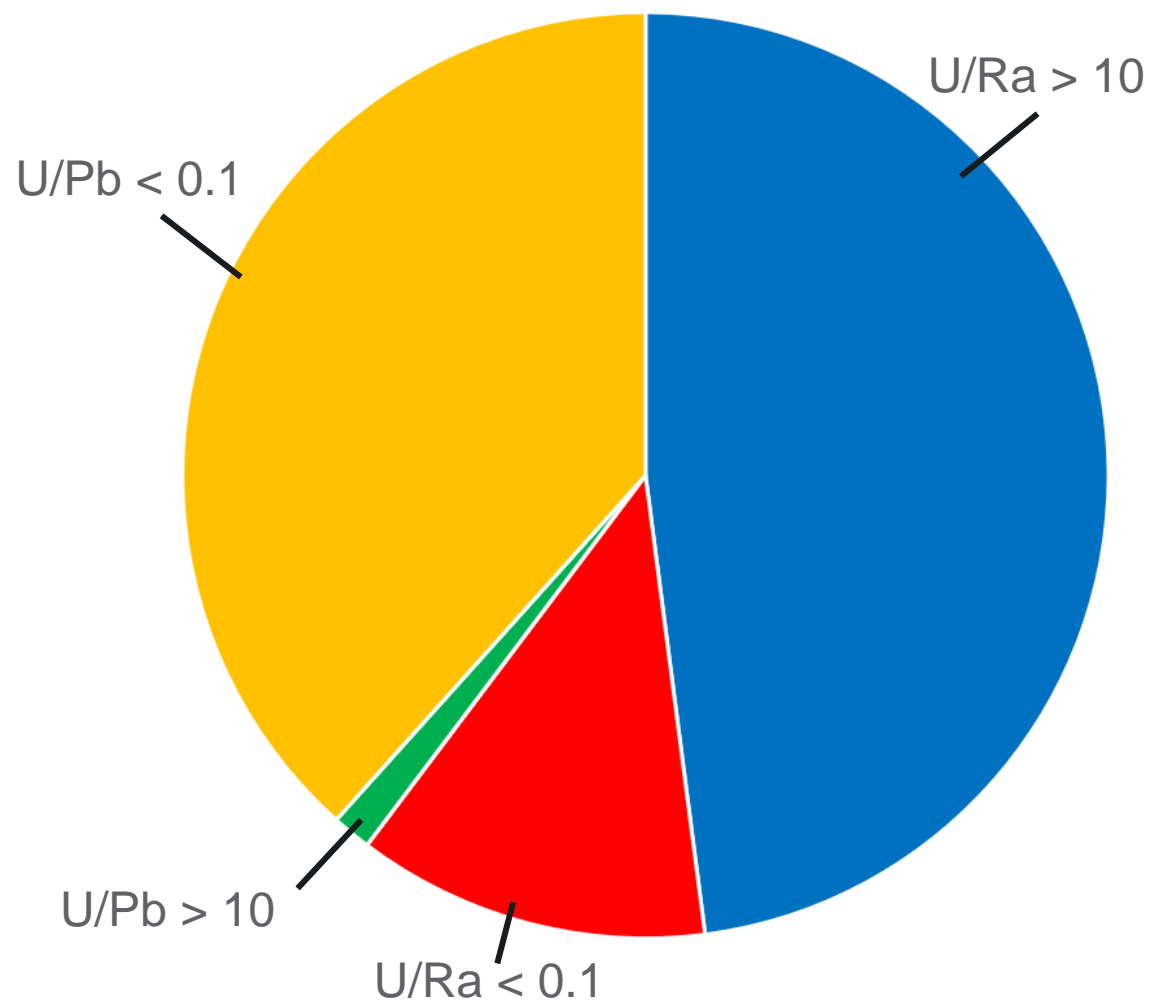
* Garnered from D.S. Akerib et al. 2020

LUX-Zeplin (LZ) Assay Data - ^{210}Pb

Samples with ^{210}Pb determinations:

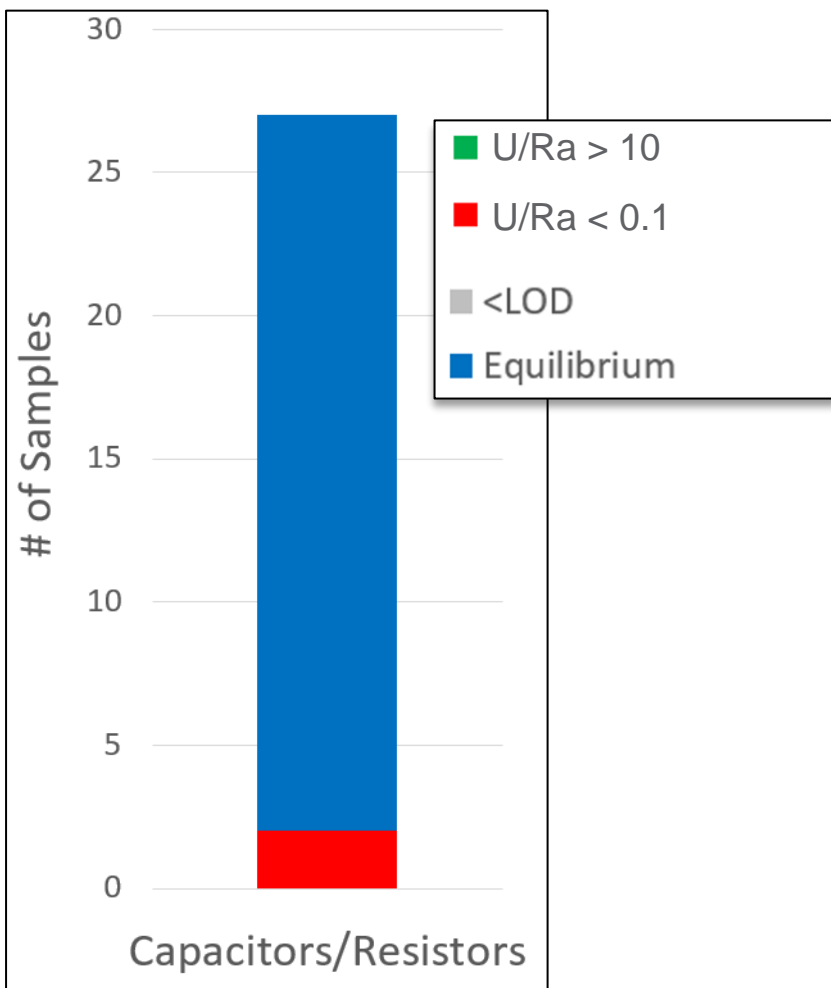


Type of disequilibrium:

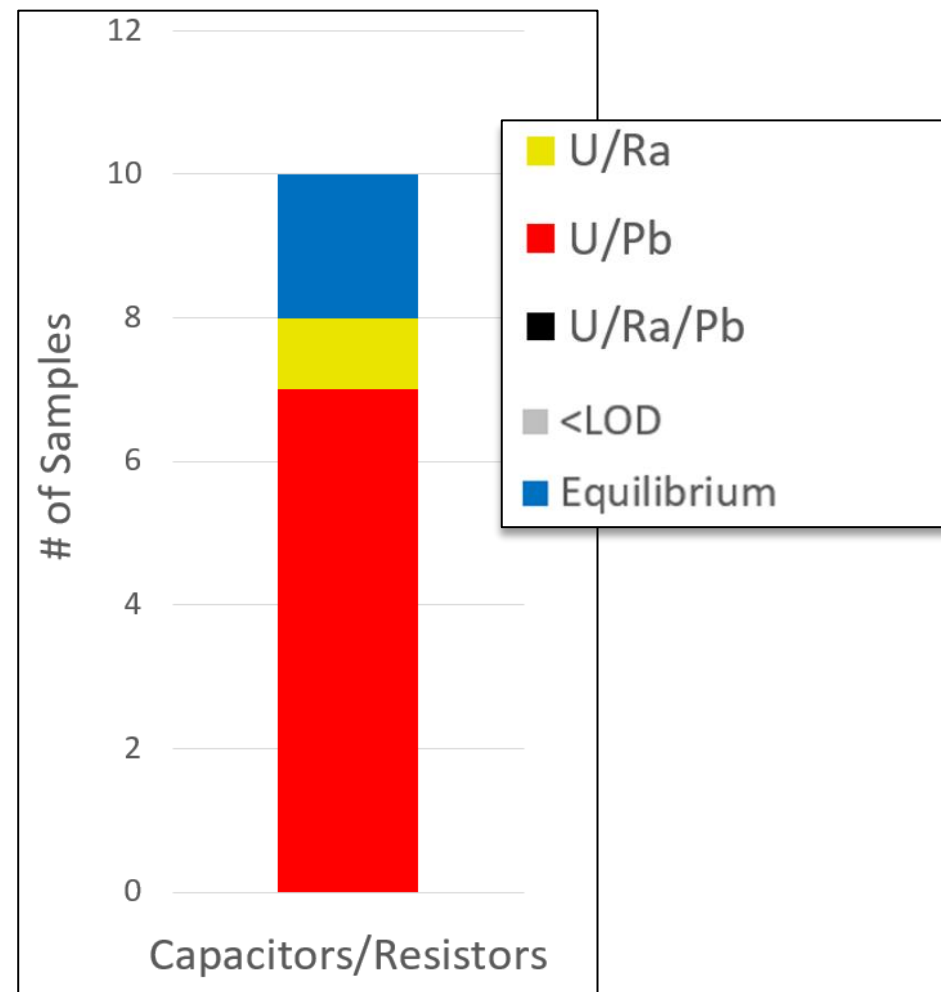


LUX-Zeplin (LZ) Assay Data – Capacitors/Resistors

U/Ra fractionation:



Samples w/ ^{210}Pb determinations:



Understanding the Problem – Better Assay

Use our automated separation method* to develop new analytical methods (measured using ICP-MS and other methods) with increased sensitivity in ^{238}U progeny assay



2DX prepFAST Station

**Automated
separation system**



How much copper is required, ideally, to attain fractions of $\mu\text{Bq/kg}$ sensitivities?



1 g
for ^{238}U



10000 g
for ^{226}Ra or ^{210}Pb

Increased sensitivity in decay chain progeny assay would allow assessment of secular disequilibria in HEP detector materials

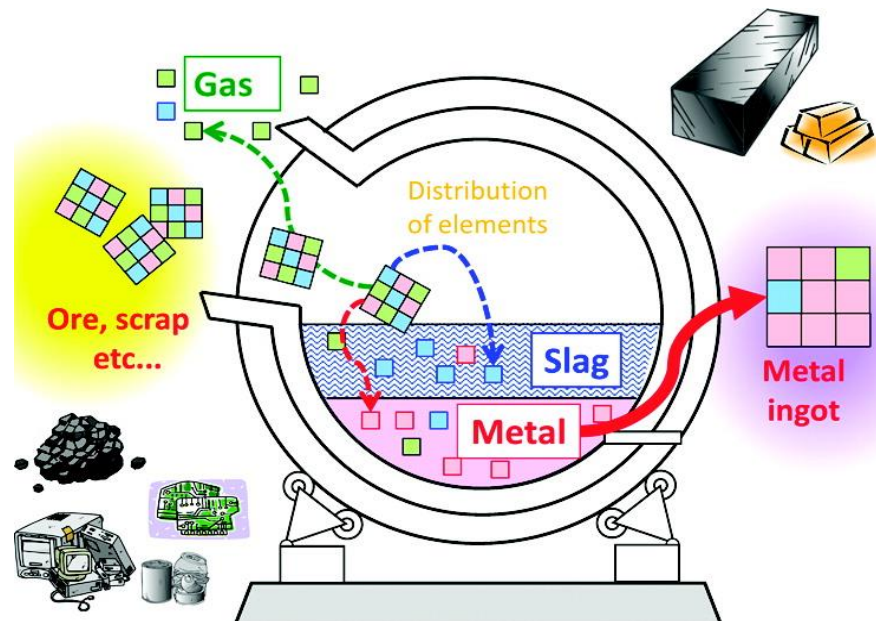
Approaches to Understand the Problem

- Spike samples with elements of interest (e.g., U, Th, ^{stable}Pb, K, etc) or other surrogate isotope for short-lived radioisotopes of interest (e.g., for Ra using Ba, which is nearly chemically identical) to track these elements through critical manufacturing steps (e.g., smelting, plating, deposition *etc.*).

1 H Hydrogen																	2 He Helium				
3 Li Lithium	4 Be Beryllium															5 B Boron	6 C Carbon	7 N Nitrogen	8 O Oxygen	9 F Fluorine	10 Ne Neon
11 Na Sodium	12 Mg Magnesium															13 Al Aluminium	14 Si Silicon	15 P Phosphorus	16 S Sulfur	17 Cl Chlorine	18 Ar Argon
19 K Potassium	20 Ca Calcium	21 Sc Scandium	22 Ti Titanium	23 V Vanadium	24 Cr Chromium	25 Mn Manganese	26 Fe Iron	27 Co Cobalt	28 Ni Nickel	29 Cu Copper	30 Zn Zinc	31 Ga Gallium	32 Ge Germanium	33 As Arsenic	34 Se Selenium	35 Br Bromine	36 Kr Krypton				
37 Rb Rubidium	38 Sr Strontium	39 Y Yttrium	40 Zr Zirconium	41 Nb Niobium	42 Mo Molybdenum	43 Tc Technetium	44 Ru Ruthenium	45 Rh Rhodium	46 Pd Palladium	47 Ag Silver	48 Cd Cadmium	49 In Indium	50 Sn Tin	51 Sb Antimony	52 Te Tellurium	53 I Iodine	54 Xe Xenon				
55 Cs Caesium	56 Ba Barium	57-71 Lanthanoids*		72 Hf Hafnium	73 Ta Tantalum	74 W Tungsten	75 Re Rhenium	76 Os Osmium	77 Ir Iridium	78 Pt Platinum	79 Au Gold	80 Hg Mercury	81 Tl Thallium	82 Pb Lead	83 Bi Bismuth	84 Po Polonium	85 At Astatine	86 Rn Radon			
87 Fr Francium	88 Ra Radium	89-103 Actinoids**		104 Rf Rutherfordium	105 Db Dubnium	106 Sg Seaborgium	107 Bh Bohrium	108 Hs Hassium	109 Mt Meitnerium	110 Ds Darmstadtium	111 Rg Roentgenium	112 Cn Copernicium	113 Nh Nihonium	114 Fl Flerovium	115 Mc Moscovium	116 Lv Livermorium	117 Ts Tennessine	118 Og Oganesson			
*Lanthanoids		57 La Lanthanum	58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium					
**Actinoids		89 Ac Actinium	90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium					

Leveraging secular disequilibrium in materials

Investigate key production steps in electrochemical and metallurgical approaches to fractionate radioisotopes in materials



K. Nakajima et al., *Thermodynamic Analysis for the Controllability of Elements in the Recycling Process of Metals*. Environ. Sci. Technol. **45** (11), 4929-4936 (2011).

Metallurgical properties to leverage

Element	T_m (°C)	T_b (°C)
Cu	1085	2571
Ti	1666	3289
U	1132	4172
Th	1750	4789
Ra	700	1737
Ba	729	1897
Pb	328	1750
K	64	765



Properties to leverage in electrochemistry

Element	Ionic Species	Standard Reduction Potential (V)	Ionic Radius (pm)
Cu	Cu^{2+}	+0.34	57-73
U	UO_2^{2+}	-1.44	~350
Th	Th^{4+}	-1.90	94-121
Ra	Ra^{2+}	-2.8	148-170
Ba	Ba^{2+}	-2.91	135-161
Pb	Pb^{2+}	-0.13	120-150
K	K^+	-2.93	137-164

Leverage chemical properties to develop methods to favorably fractionate the decay progeny in materials ($^{238}\text{U} \gg ^{226}\text{Ra}$ and ^{210}Pb)

Summary

- ICP-MS and NAA assays measure ^{238}U at nanoBq/kg levels. However, estimating backgrounds from ^{238}U can lead to over- or under-estimated background contributions if materials are not in secular equilibrium.
- HPGe assay does not suffer from this problem but is limited to about microBq/kg sensitivities.
 - Many experiments require materials (e.g., copper, electronics components, *etc.*) in nanoBq/kg range → creates blindspots between ICP-MS/NAA and Ge
- This can be especially pertinent to materials (e.g., copper, polymers) used in high volume because, while very radiopure, they can make up the largest background contribution (e.g., nEXO, SCDMS, *etc.*).
- Secular disequilibrium in the ^{238}U series decay chain has been measured in many different materials across multiple detectors
- Several approaches are available to investigate procedures that promote ***favorable*** secular disequilibria (^{210}Pb , $^{226}\text{Ra} \ll ^{238}\text{U}$) and leverage these procedures to reduce backgrounds in ULB detectors

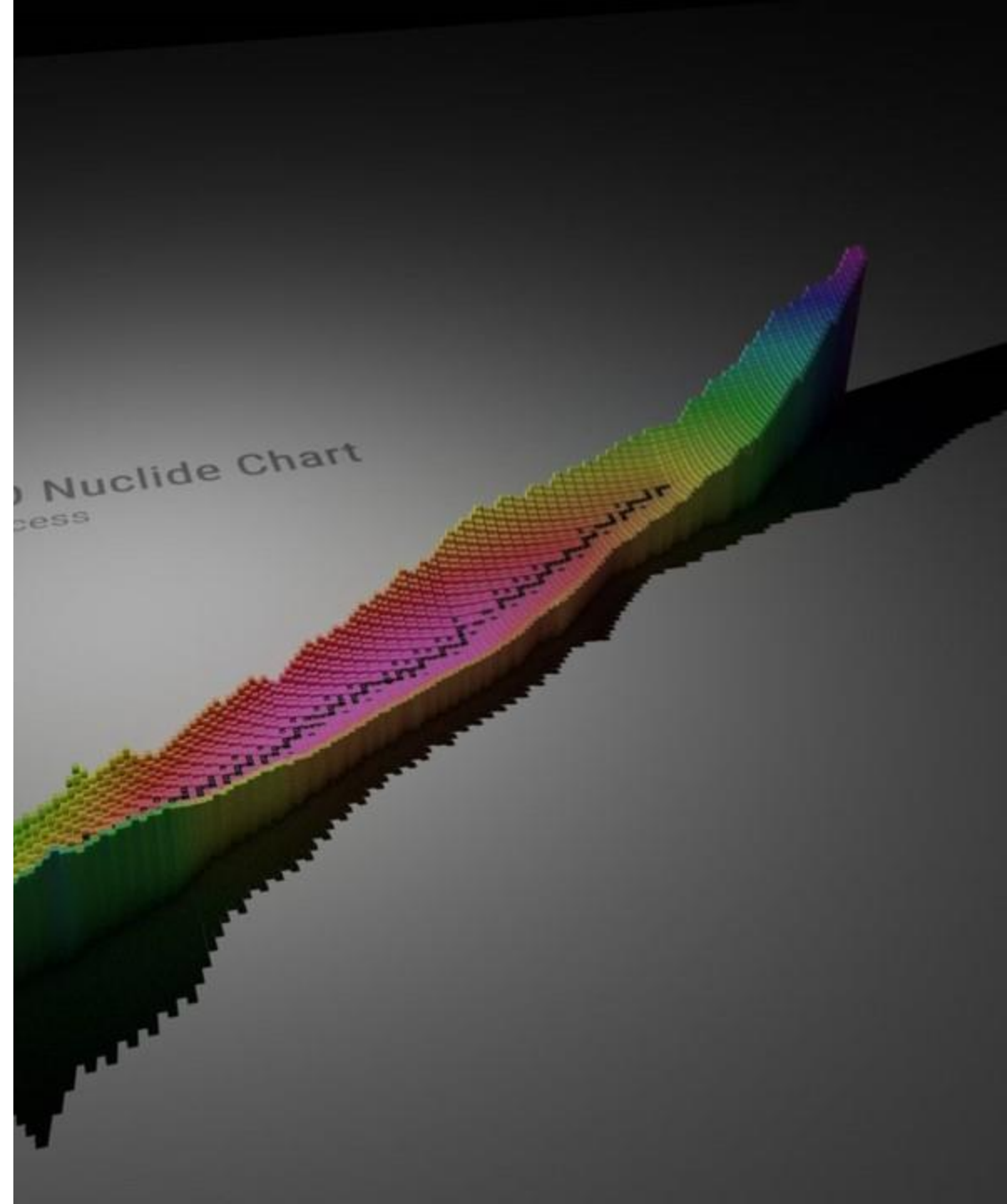


Thank you

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PNNL is operated by Battelle for the U.S. Department of Energy

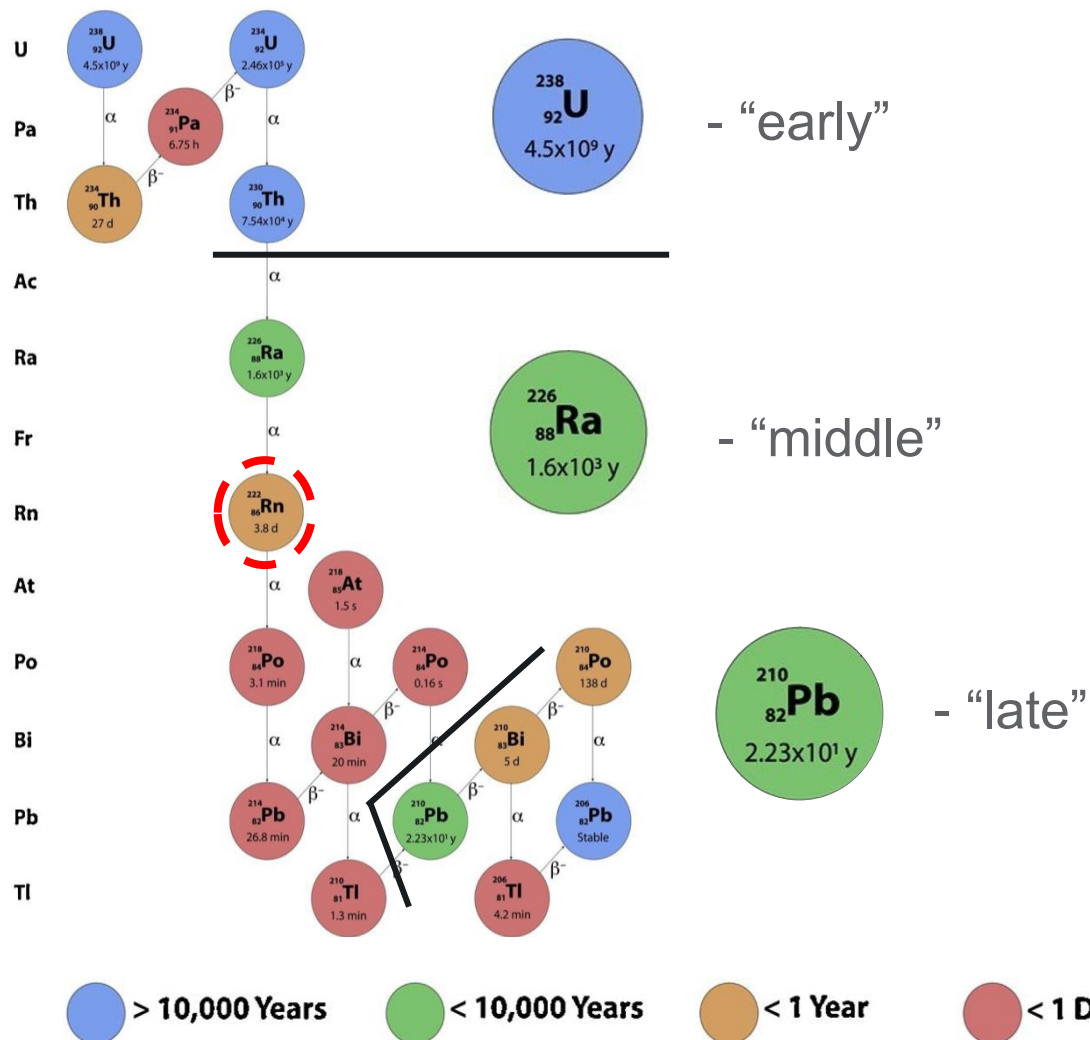


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Decay Chains

^{238}U decay chain



^{232}Th decay chain

