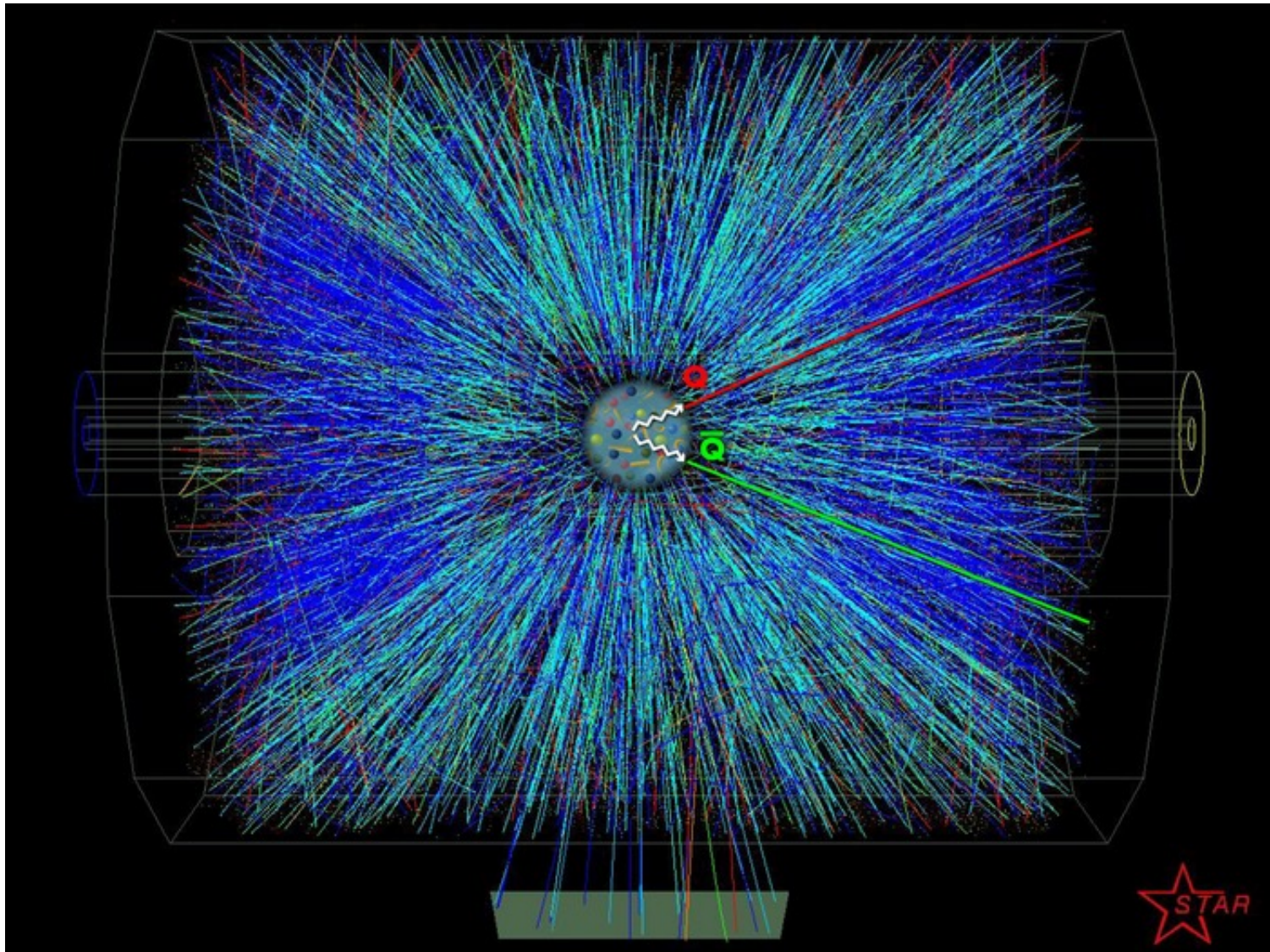
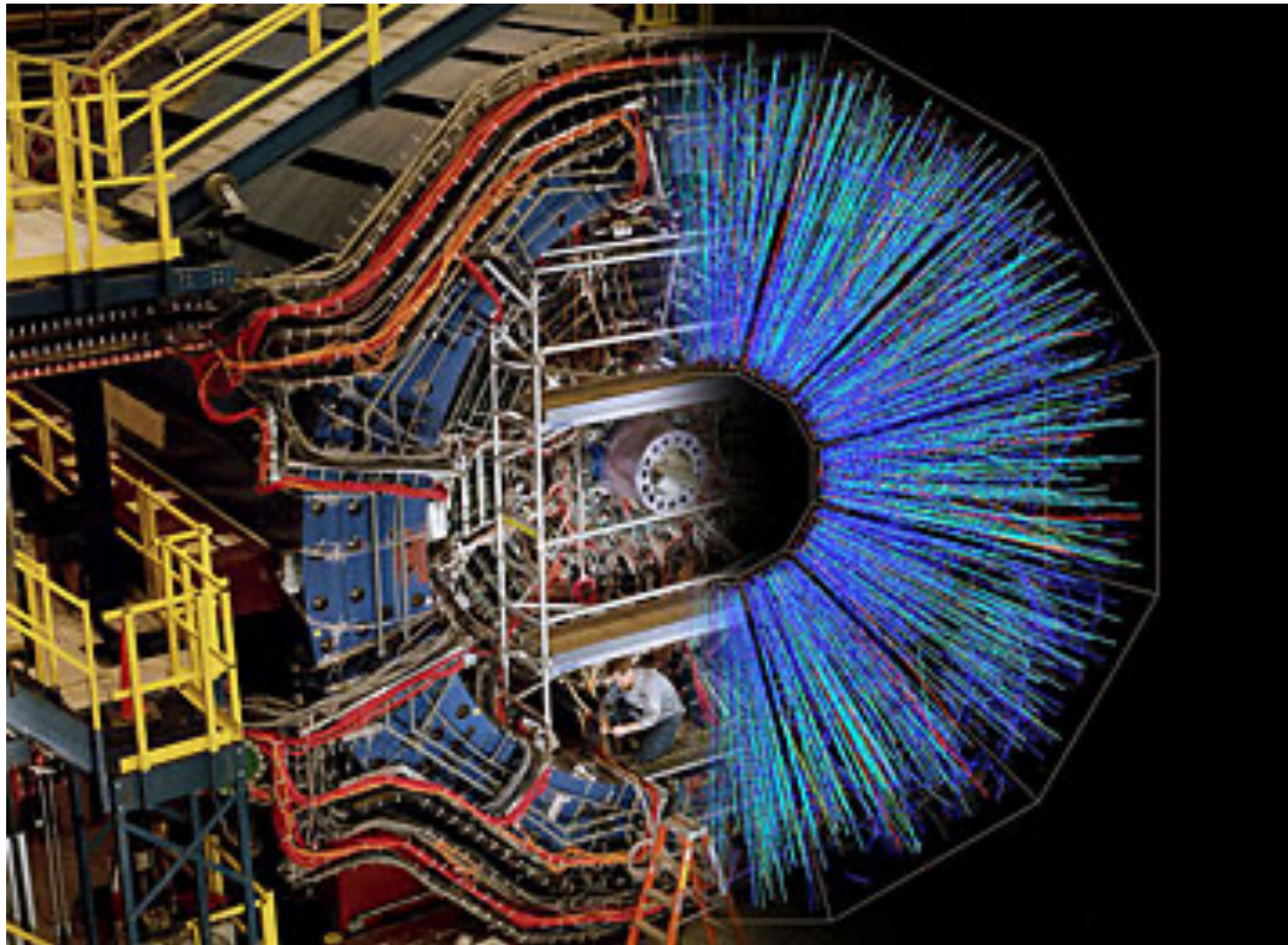


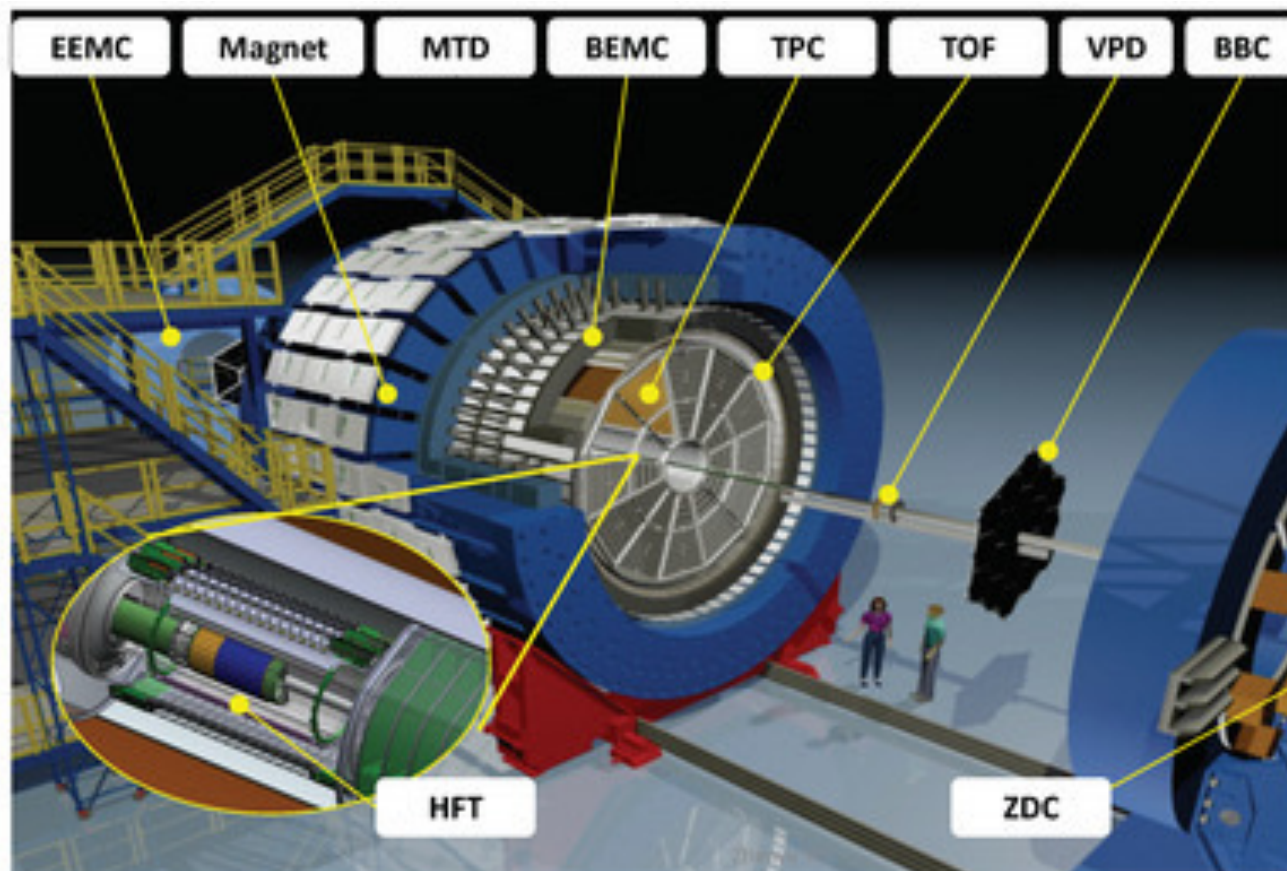
Small-Strip Thin Gap Chamber (sTGC)

Prashanth Shanmuganathan

Why we need sTGC?







- **Tracking and PID (full 2π)**

TPC: $|\eta| < 1$

TOF: $|\eta| < 1$

BEMC: $|\eta| < 1$

EEMC: $1 < \eta < 2$

HFT (2014-2016): $|\eta| < 1$

MTD (2014+): $|\eta| < 0.5$

- **MB trigger and event plane reconstruction**

BBC: $3.3 < |\eta| < 5$

EPD (2018+): $2.1 < |\eta| < 5.1$

FMS: $2.5 < \eta < 4$

VPD: $4.2 < |\eta| < 5$

ZDC: $6.5 < |\eta| < 7.5$

Concept

- An infinite plane metal plate is in the xy -plane. A point charge $+Q$ is placed on the z -axis at a height h above the plate. Consequently, electrons will be attracted to the part of the plate immediately below the charge, so that the plate will carry a negative charge density σ which is greatest at the origin and which falls off with distance ρ from the origin.

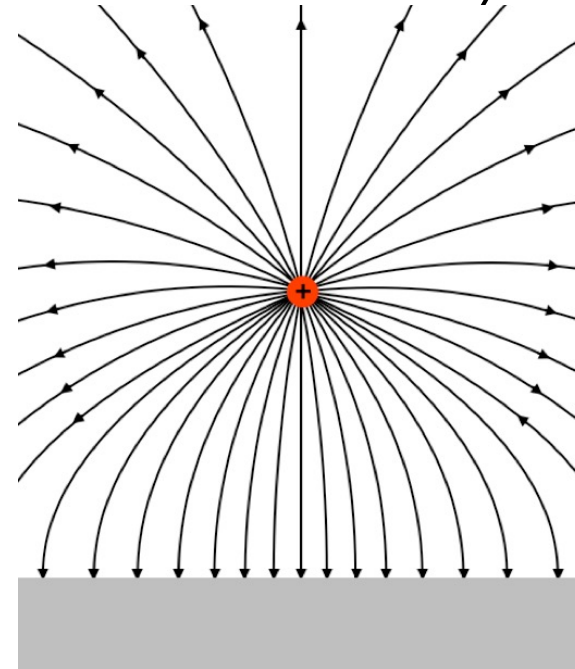
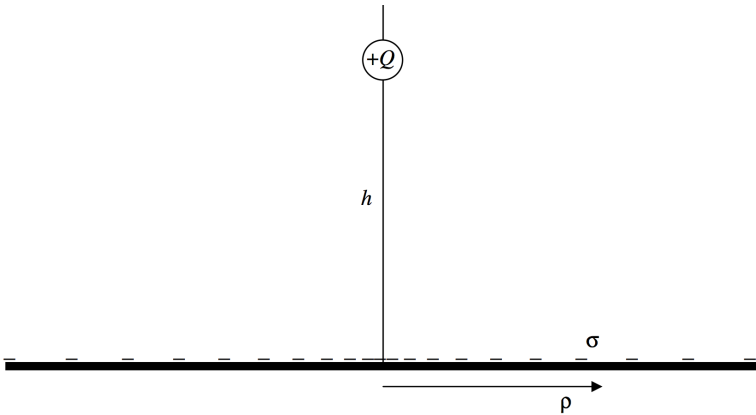
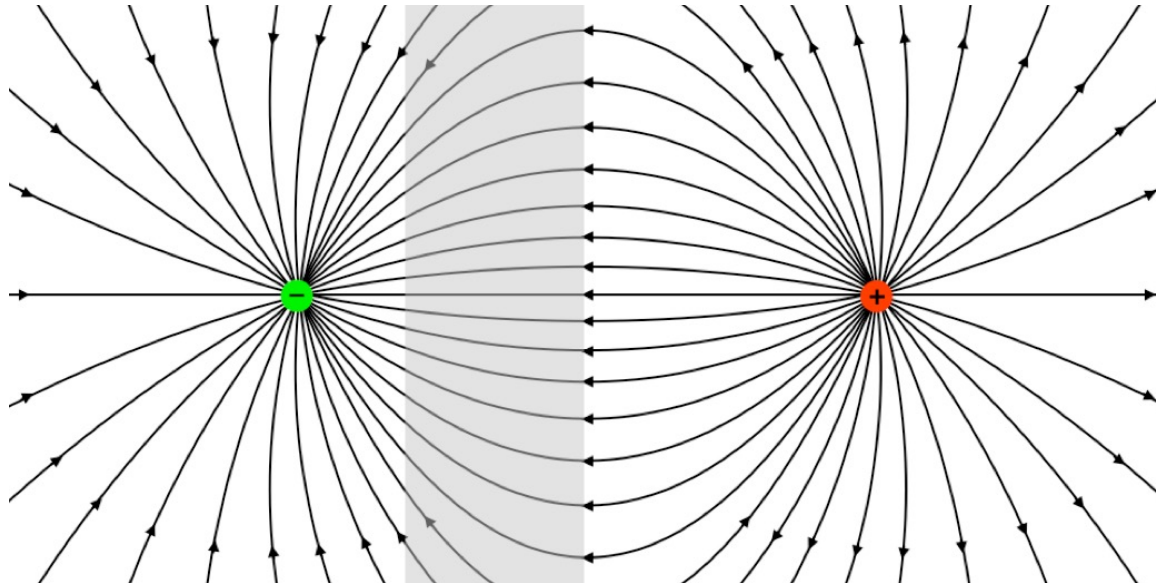


Image Charge

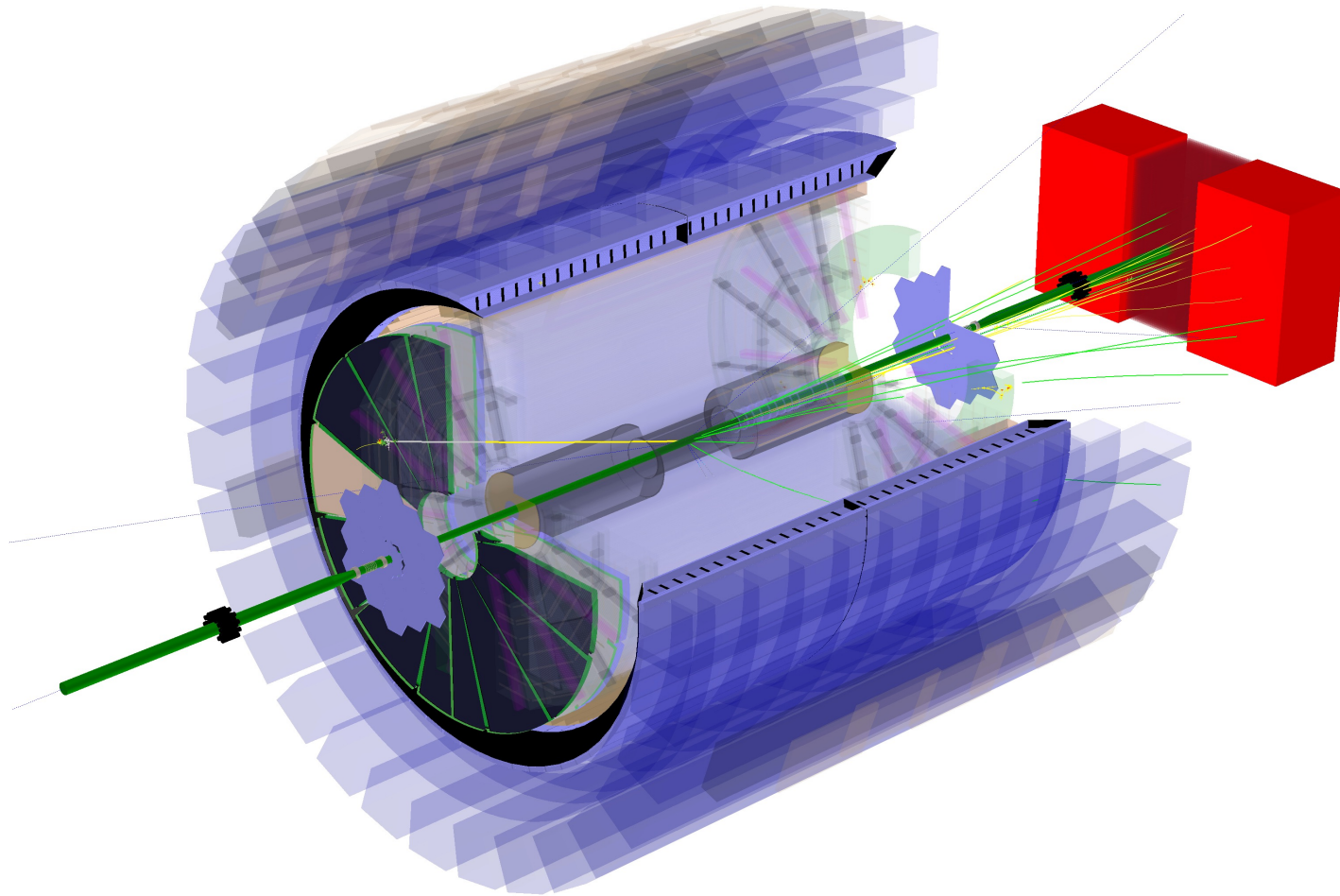


A Clever Trick

The equivalence of these two fields provides us with an opportunity to use a clever trick for analyzing physical situations involving electric charges near flat conductors. For a point charge, this trick involves introducing an imaginary *image charge* reflected across the conducting surface, and using that charge to derive the actual field outside the conductor surface.

Alert

It can't be stressed enough that this trick does not involve introducing an actual physical charge, any more than constructing a gaussian surface involved constructing an actual physical surface. These are techniques for performing calculations, and one should always keep in mind what the actual physical circumstances are.



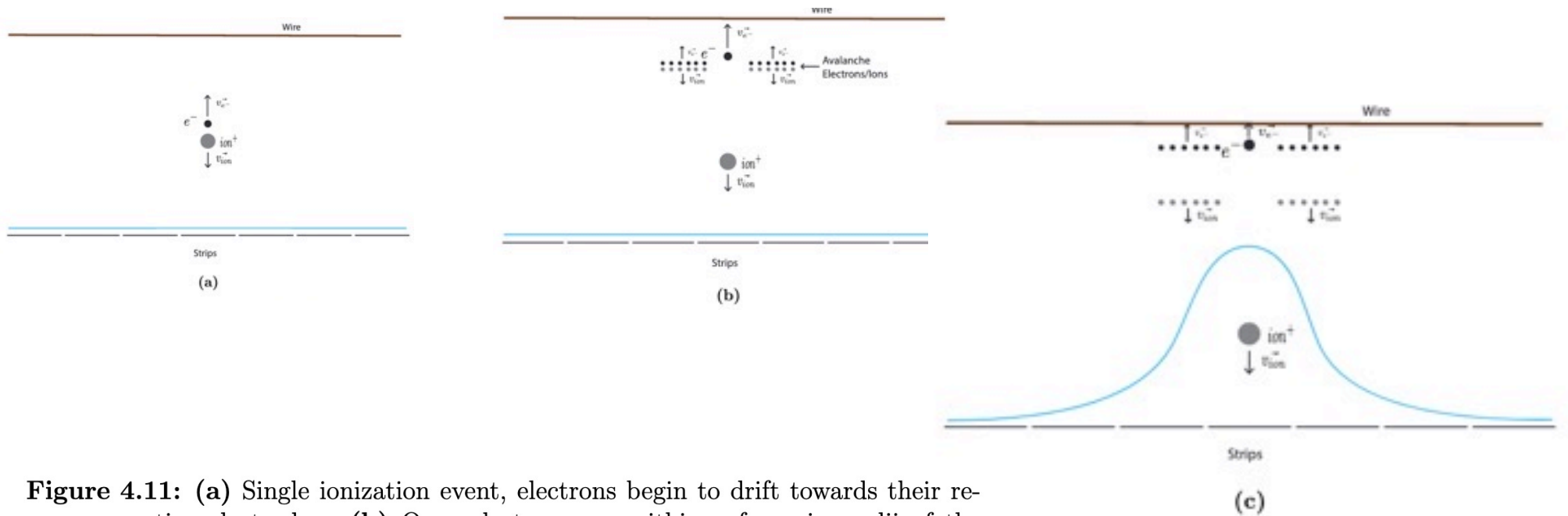


Figure 4.11: (a) Single ionization event, electrons begin to drift towards their respective electrodes. (b) Once electrons are within a few wire radii of the wire they have enough energy to produce secondary ionization electrons in an electron avalanche. (c) The rapid production of electrons at the wire induces charge on the strips. The charge density will follow a Gaussian-like distribution over the strips. The blue line depicts the charge distribution when the relative amplitudes of the strip signals are plotted.

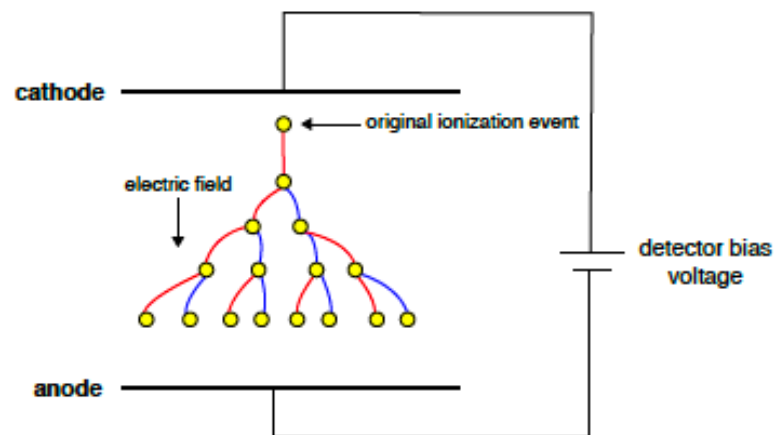


Figure 2.6: Schematic diagram of a Townsend avalanche. A strong electric field permeates the gaseous medium between the anode and the cathode. The yellow disks represent ionization events caused by an incident electron (red lines). Freed electrons (blue lines) produce more ionization events as they drift in the electric field. The total ionization approximately doubles at each step of the avalanche.

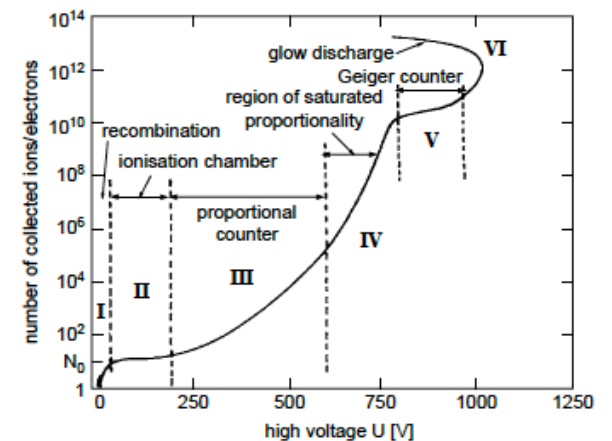
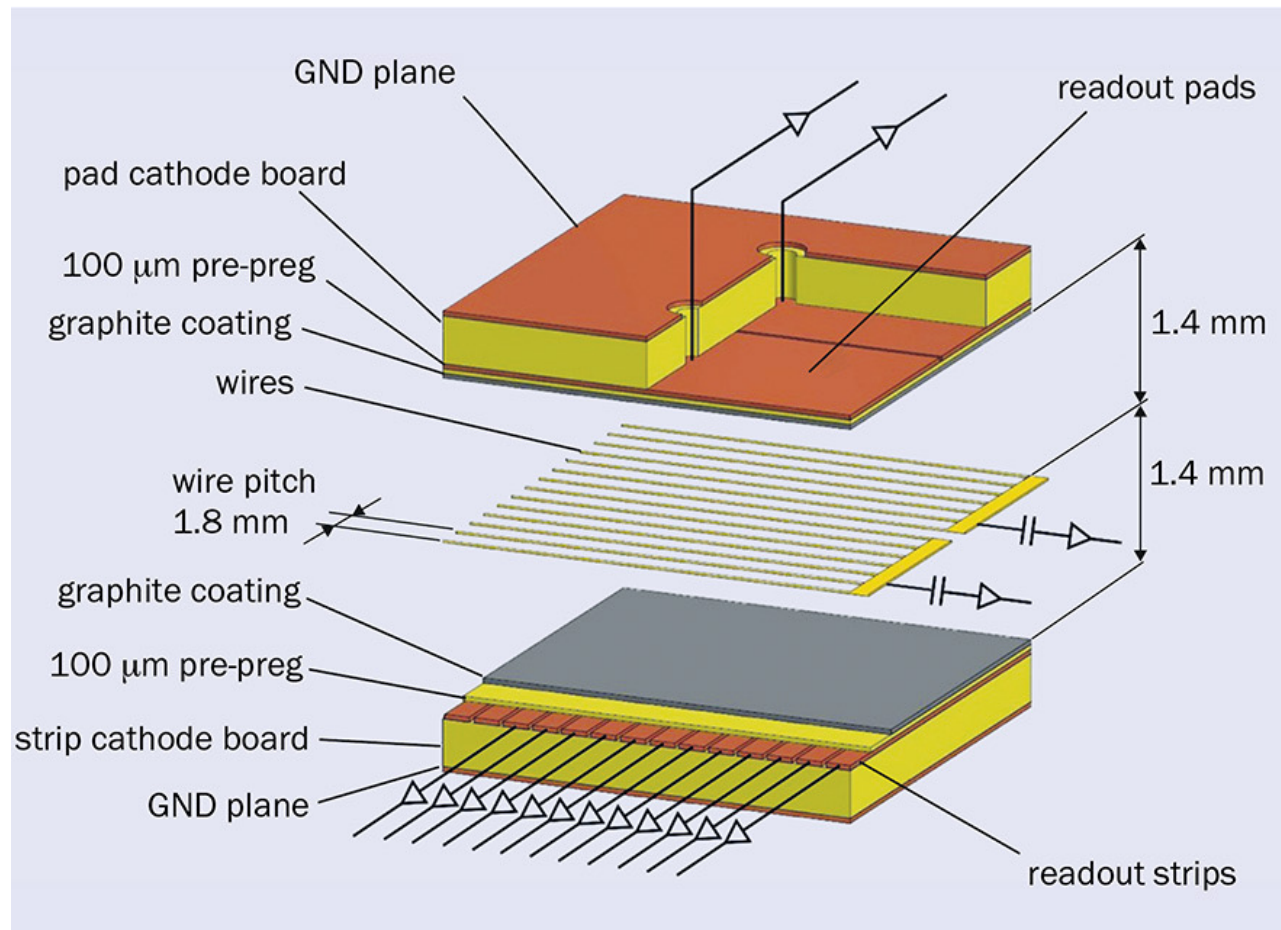


Figure 2.7: Ionization yield as a function of voltage for a typical ionization detector following the passage of a mip with a primary ionization $N_0 \sim 10$ [32].

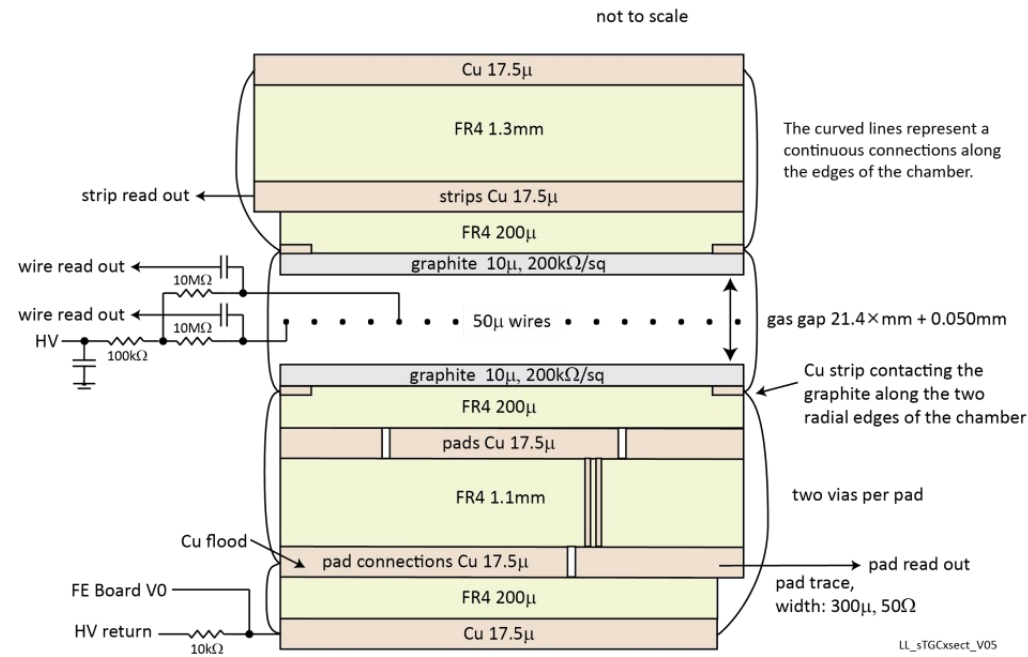
Concept to Reality

sTGC



Detailed cryosections of sTGC

sTGC cross-section



sTGC

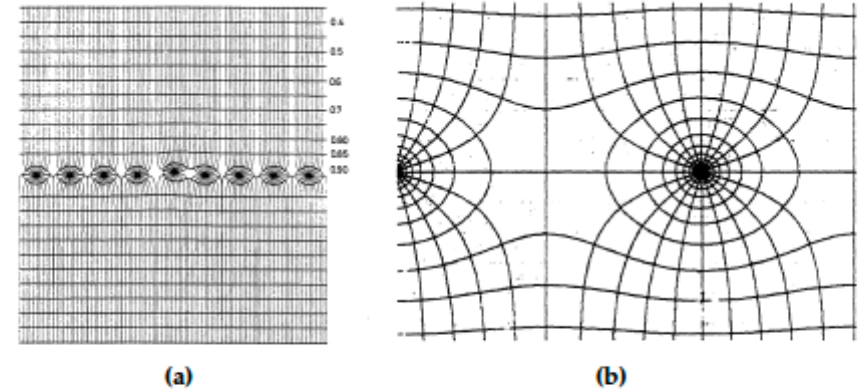
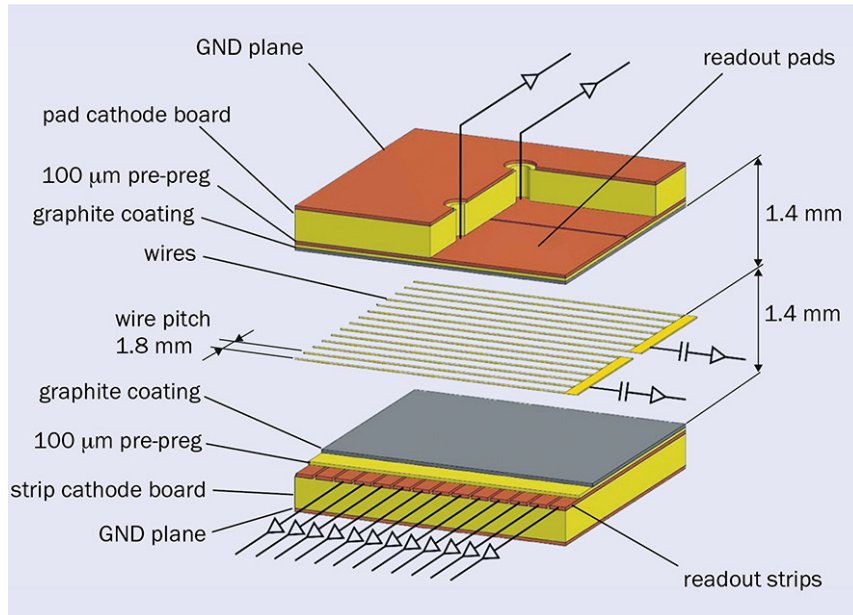
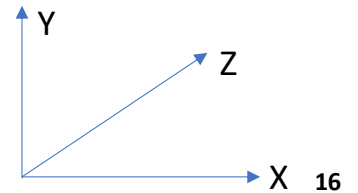
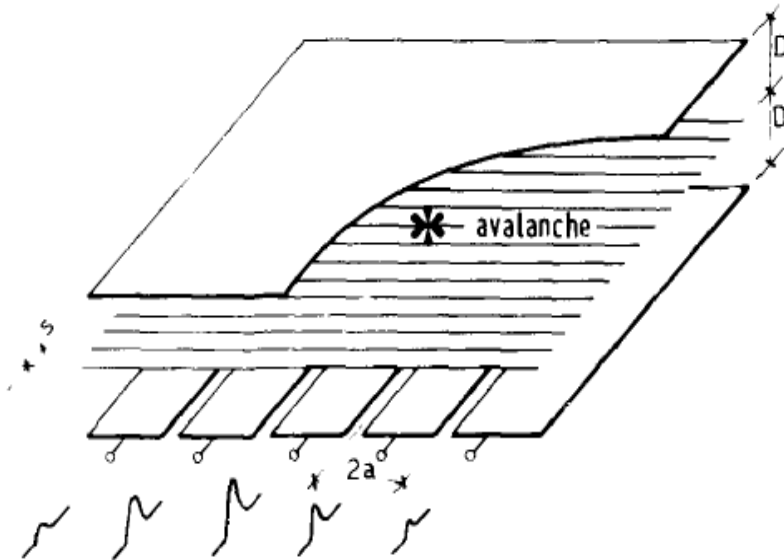
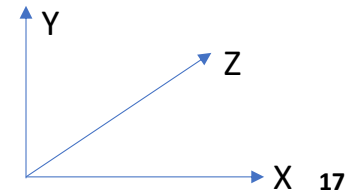
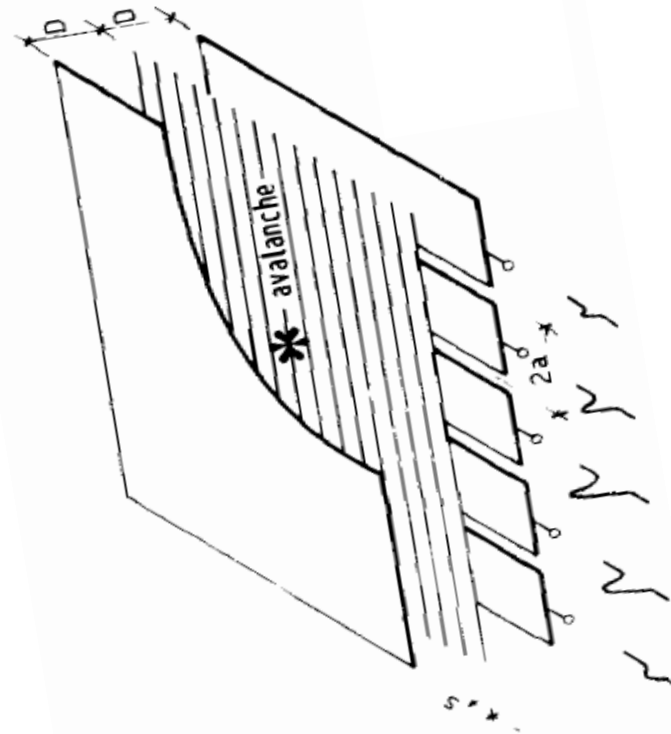
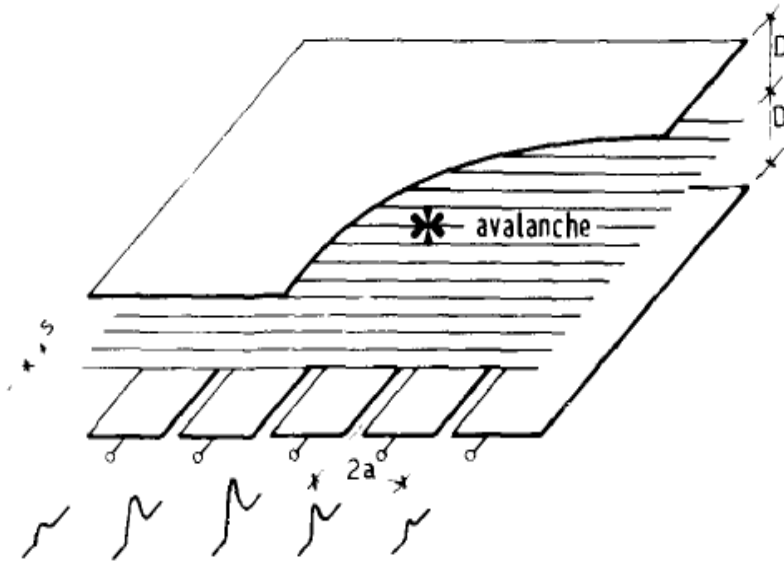


Figure 2.11: (a) Electric field lines and equipotentials in the gas volume of a multi-wire chamber. The electric potential relative to the anode wire potential, or value of V_0 , is indicated. The effects of a minor displacement of one anode wire are shown. (b) Enlarged view of the same figure in the vicinity of a wire [14].

Track reconstruction – 1D positioning



Track reconstruction – 2D positioning



Choice of gas?

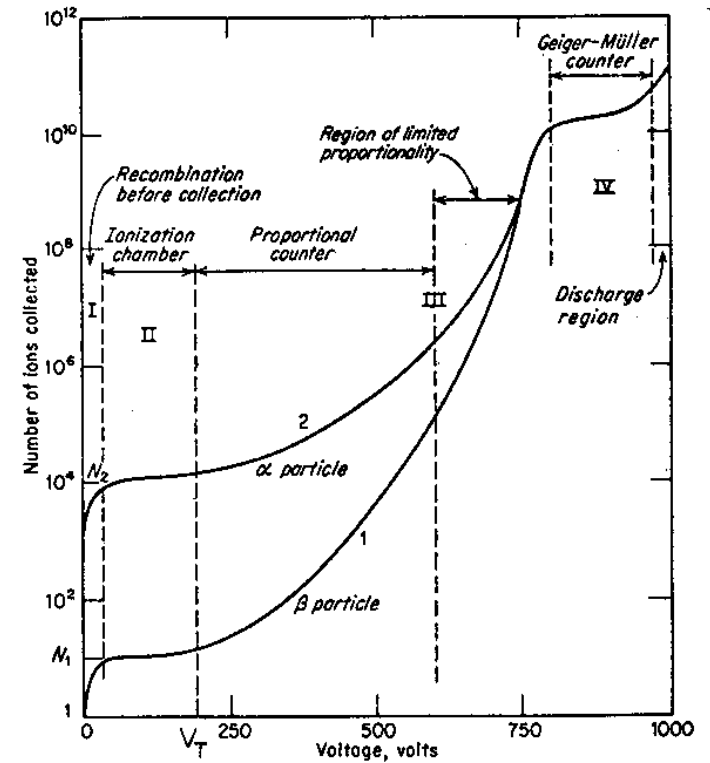
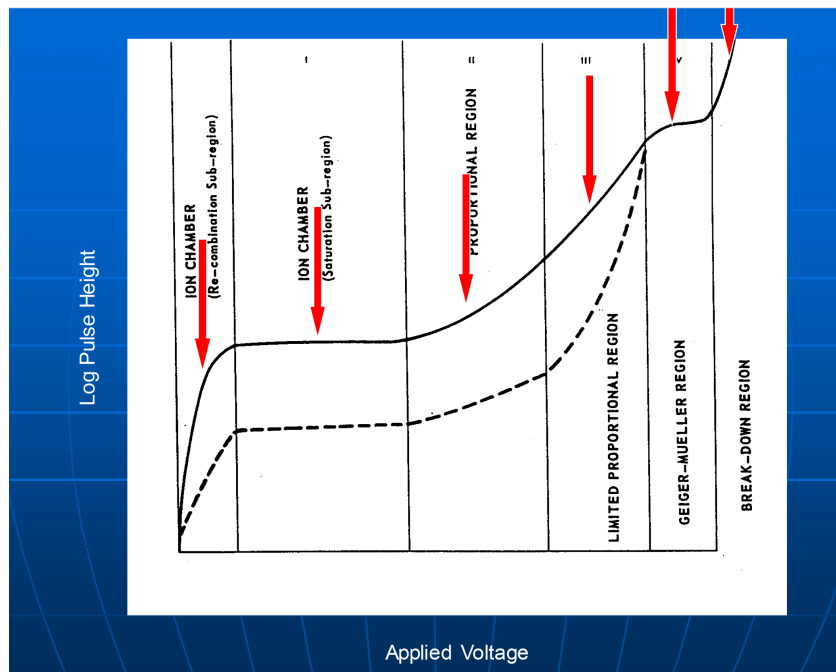


Fig. 50 Gain-voltage characteristics for a proportional counter, showing the different regions of operation (from W. Price, see bibliography for Sections 2 and 3).

Gas

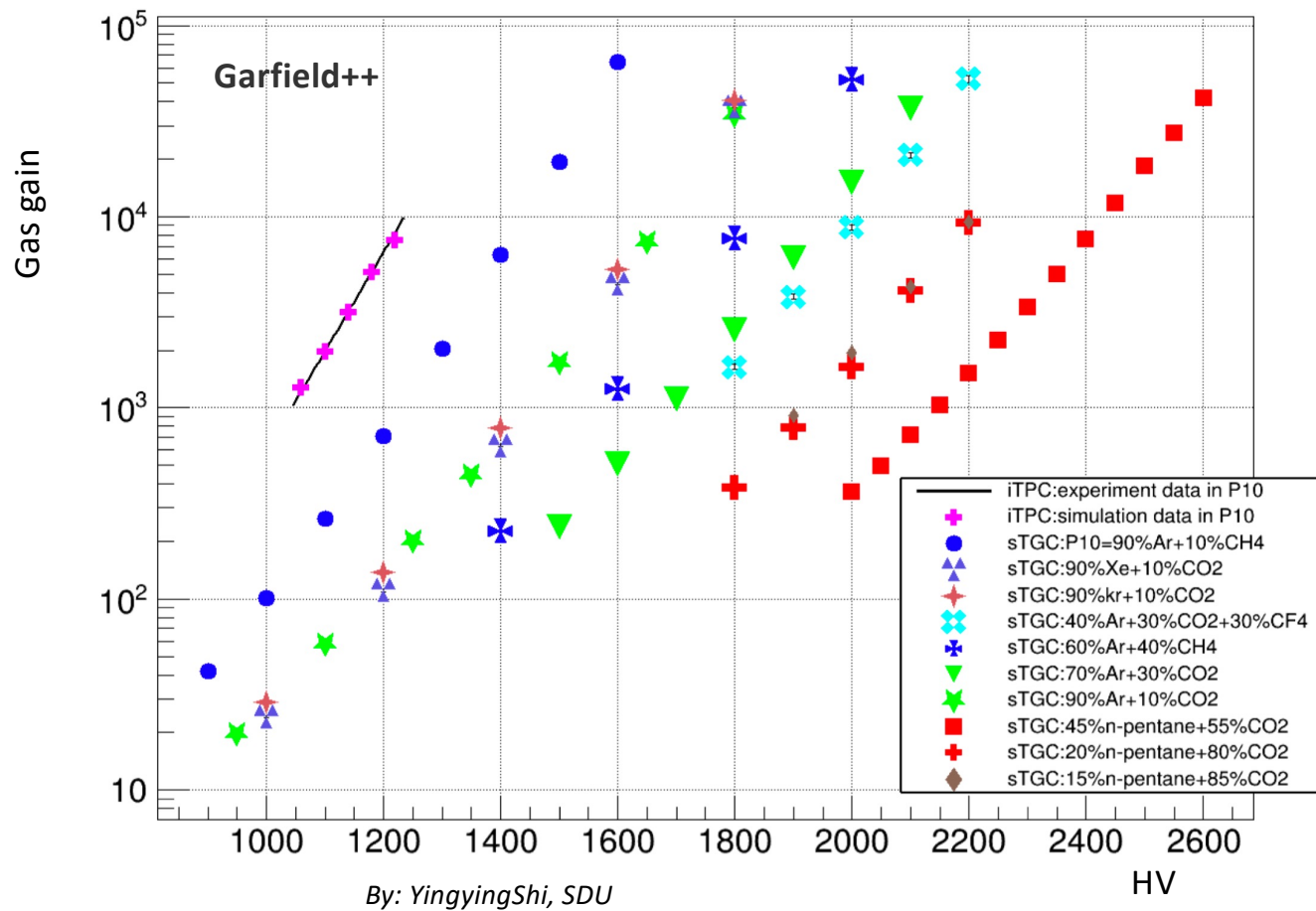
Table 1

Properties of several gases used in proportional counters (from different sources, see the bibliography for this section). Energy loss and ion pairs per unit length are given at atmospheric pressure for minimum ionizing particles

Gas	Z	A	δ (g/cm ³)	E _{ex}	E _i (eV)	I ₀	W _i	dE/dx		n _p (i.p./cm) ^{a)}	n _T (i.p./cm) ^{a)}
								(MeV/g cm ⁻²)	(keV/cm)		
H ₂	2	2	8.38×10^{-5}	10.8	15.9	15.4	37	4.03	0.34	5.2	9.2
He	2	4	1.66×10^{-4}	19.8	24.5	24.6	41	1.94	0.32	5.9	7.8
N ₂	14	28	1.17×10^{-3}	8.1	16.7	15.5	35	1.68	1.96	(10)	56
O ₂	16	32	1.33×10^{-3}	7.9	12.8	12.2	31	1.69	2.26	22	73
Ne	10	20.2	8.39×10^{-4}	16.6	21.5	21.6	36	1.68	1.41	12	39
Ar	18	39.9	1.66×10^{-3}	11.6	15.7	15.8	26	1.47	2.44	29.4	94
Kr	36	83.8	3.49×10^{-3}	10.0	13.9	14.0	24	1.32	4.60	(22)	192
Xe	54	131.3	5.49×10^{-3}	8.4	12.1	12.1	22	1.23	6.76	44	307
CO ₂	22	44	1.86×10^{-3}	5.2	13.7	13.7	33	1.62	3.01	(34)	91
Cl ₄	10	16	6.70×10^{-4}		15.2	13.1	28	2.21	1.48	16	53
C ₄ H ₁₀	34	58	2.42×10^{-3}		10.6	10.8	23	1.86	4.50	(46)	195

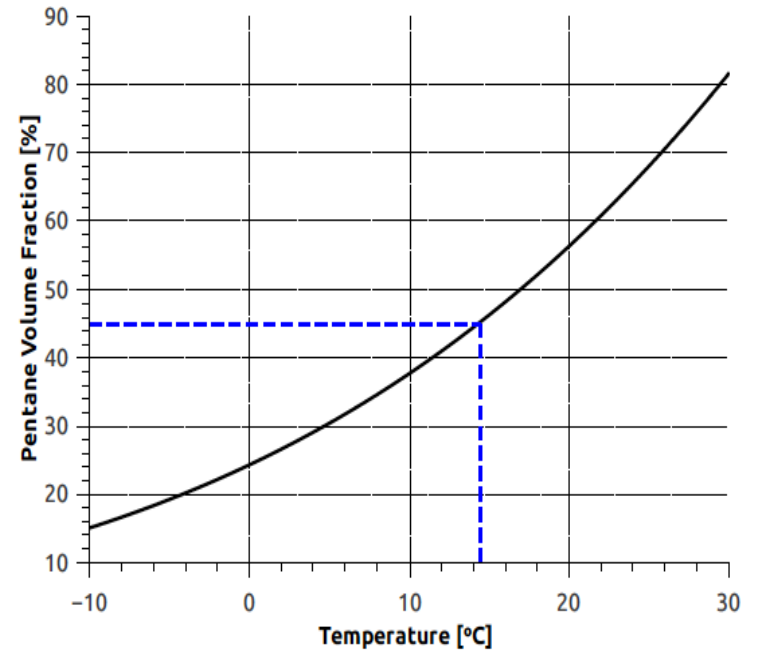
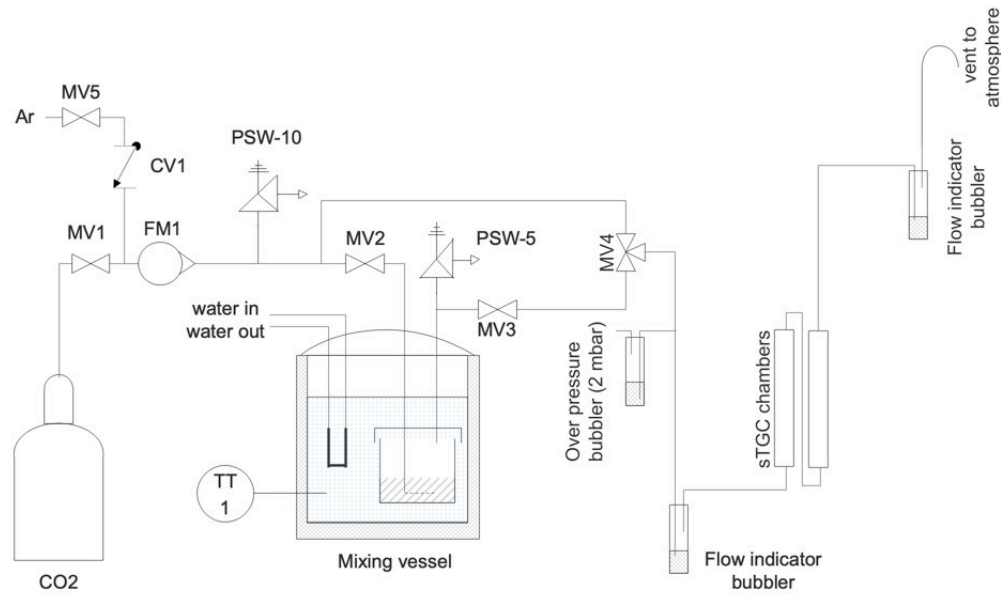
a) i.p. = ion pairs

Gas Choices



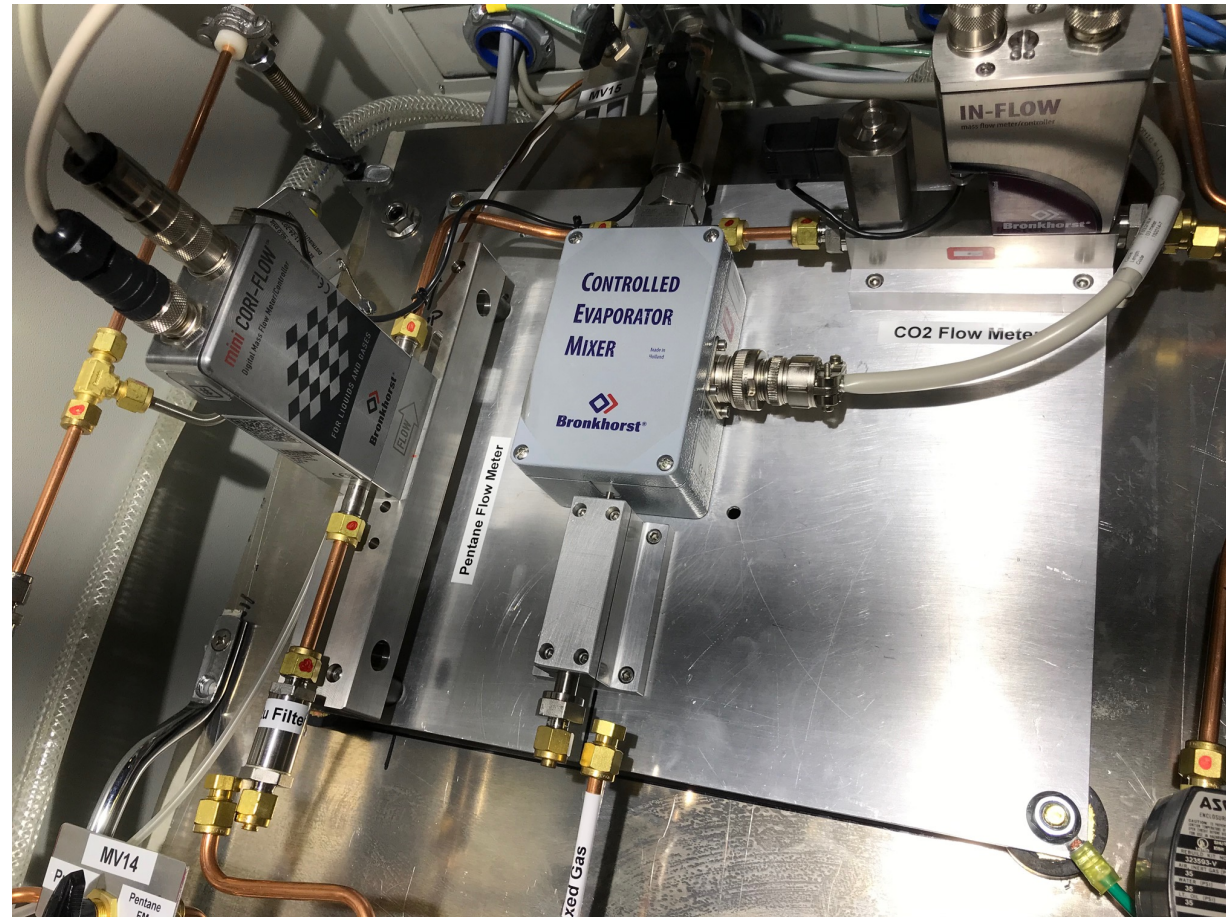
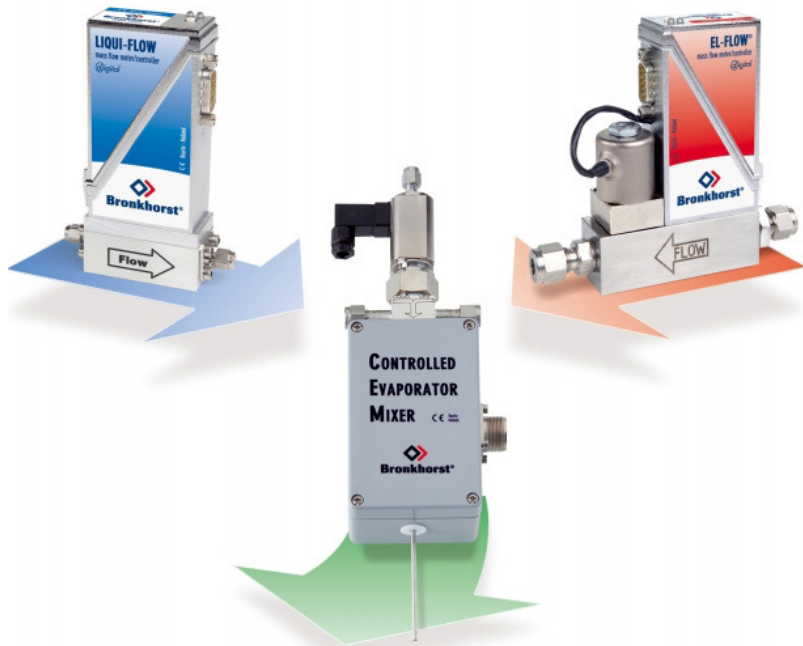
- **Quenching gas is essential for stable operation:** sTGCs operate at high gas amplification, meaning a small initial ionization from a particle interaction is greatly amplified to produce a detectable signal. Without a quenching gas, this amplification process could lead to uncontrolled discharges (streamers or sparks) and damage the detector.
- **The gas mixture acts as a quenching agent:** The operational gas is typically a mixture of **55% CO₂ and 45% n-pentane**. The n-pentane acts as a quenching gas, absorbing photons and ions produced during the electron avalanche and preventing them from initiating further avalanches.
- **Benefits of using this specific mixture:** This gas mixture allows sTGCs to operate at high gas amplification while maintaining stability and minimizing the risk of discharges. Tests have shown that sTGCs operating with this mixture exhibit greater stability and lower sensitivity to electric field non-uniformity compared to using only CO₂.
- **Significance in High-Luminosity Environments:** The ability of this gas mixture to ensure stable operation at high gas amplification is crucial for detectors like sTGCs designed to function in the high-radiation environment of the High-Luminosity LHC.

Getting the right mixture



arXiv:1702.01240v3 [physics.ins-det]

Getting the right mixture



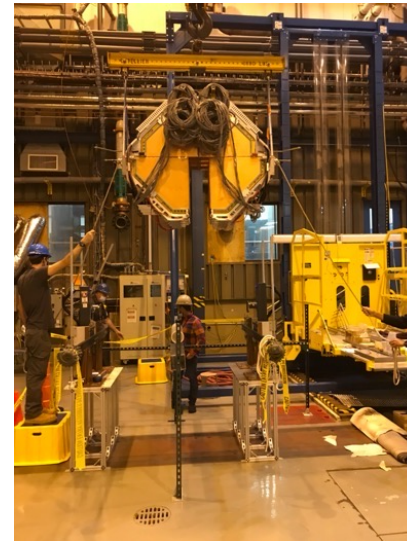
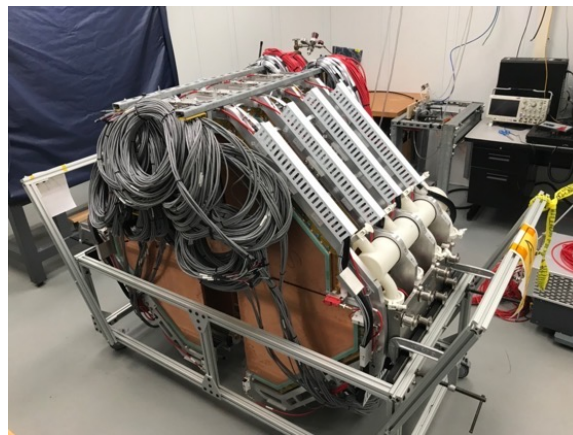
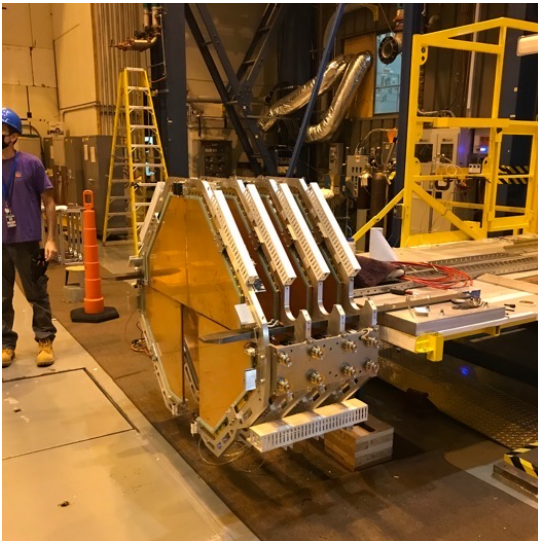
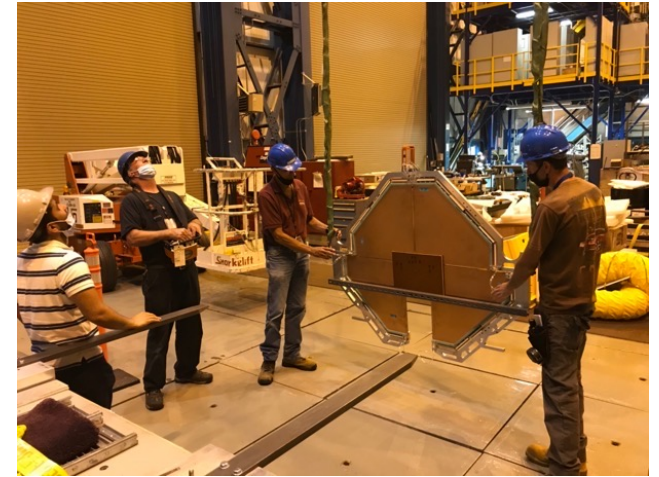
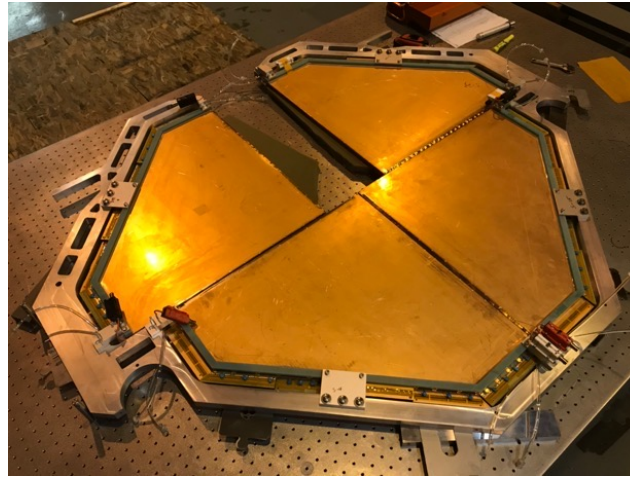
Bronkhorst components assembled in the gas cabinet

sTGC Operations Reequipments

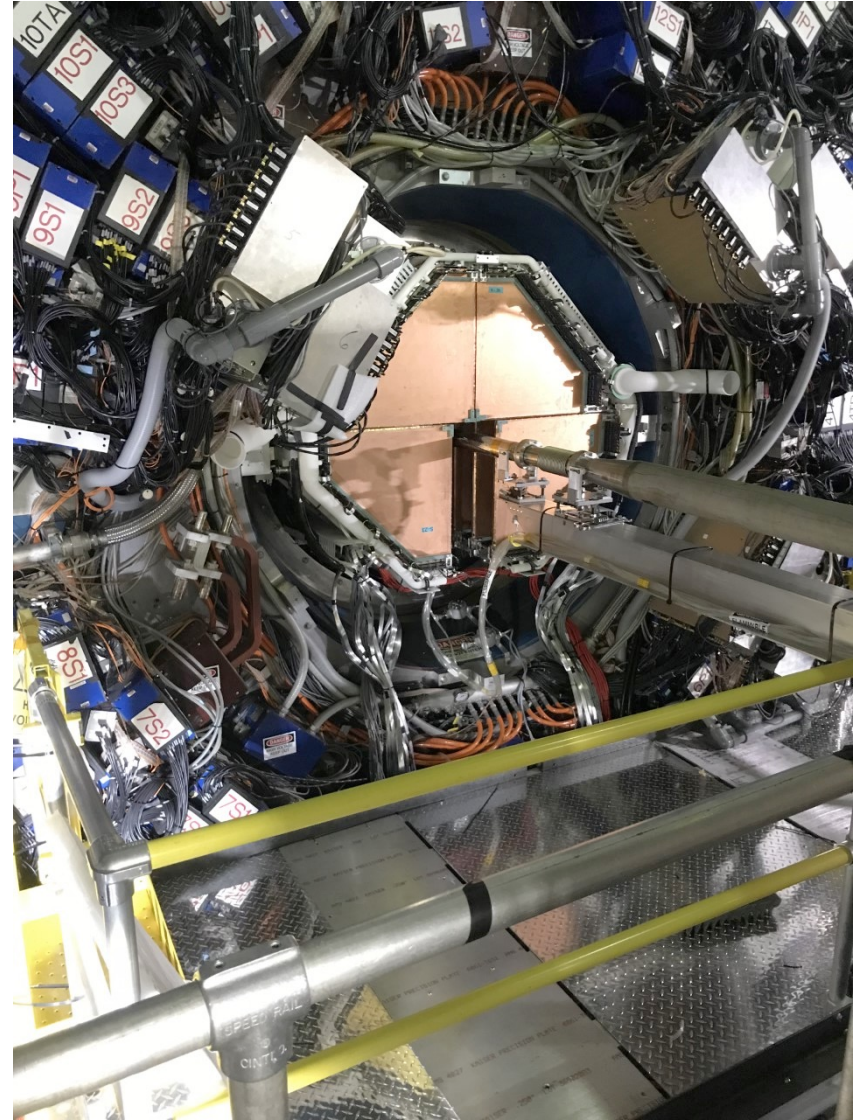
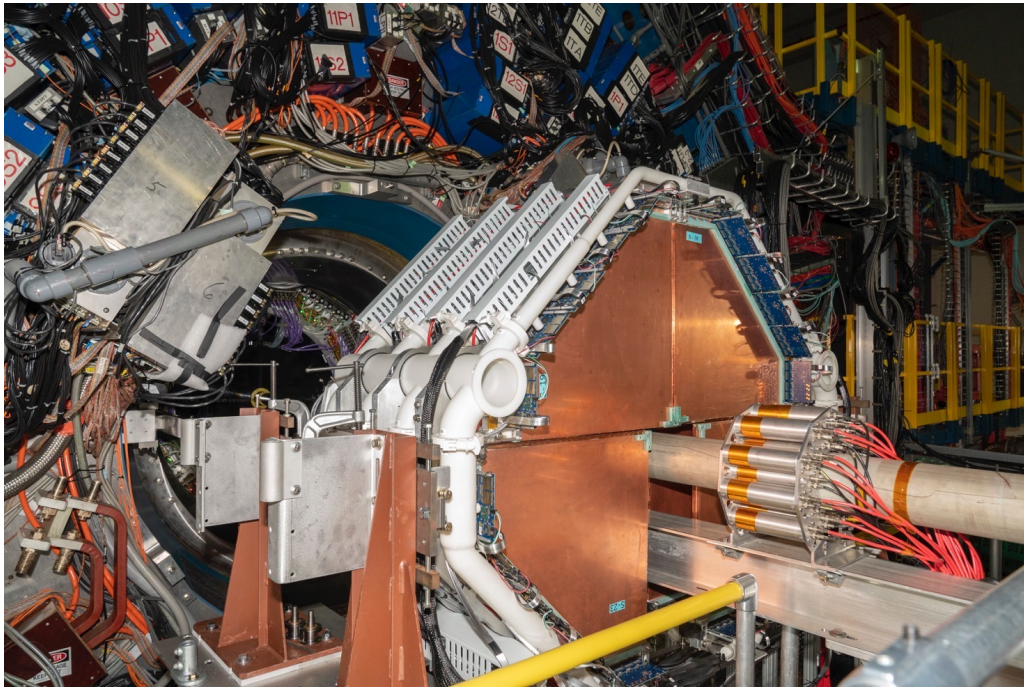
- Anode (HV): 50 μm gold-plated tungsten wires held at a potential of $\sim 2900\text{ V}$
- Working gas: n-Pentane+CO₂= 45:55% by volume
- Supply pressure 2 mbar above atm
- Flow about 50 cc/min

Why sTGC?

sTGC Detector Assembly

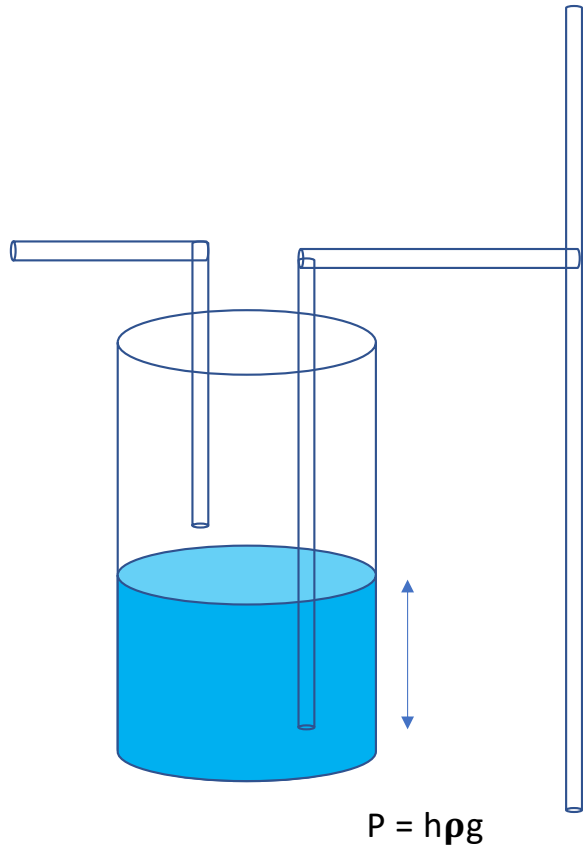


sTGC Detector Assembly

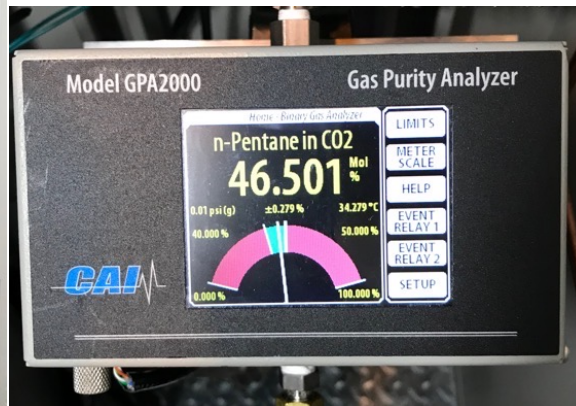
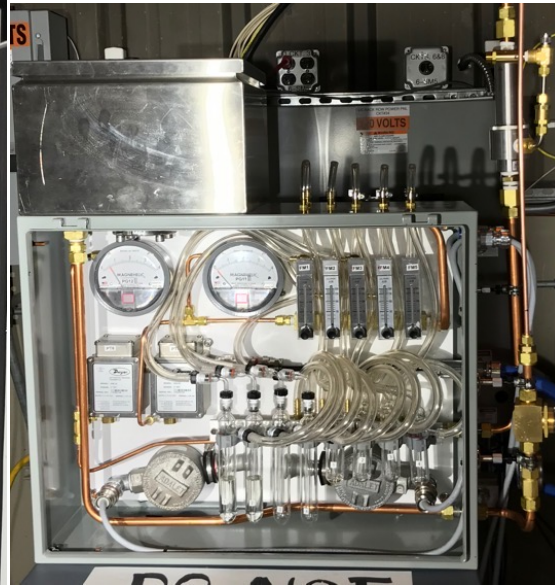


Building the Gas System

Protecting the chambers from over pressure



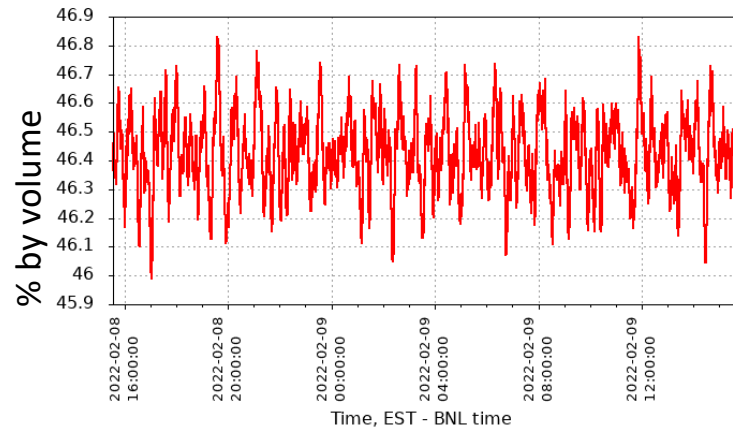
Gas System



n-Pentane

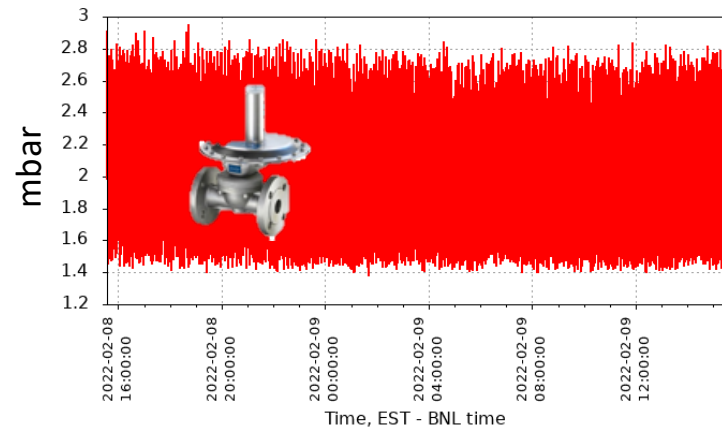
- n-pentane isomer formula C_5H_{12}
- Is a highly flammable liquid and vapor
- Boiling point of pentane is $97^{\circ}F$ ($36^{\circ}C$)
- Density of pentane is 0.626 g/ml
- The pentane vapor is heavier than air
 - It sinks if released to atmosphere
- Explosive limits of pentane by volume in air: 1.4-7.8%

sTGC Gas System



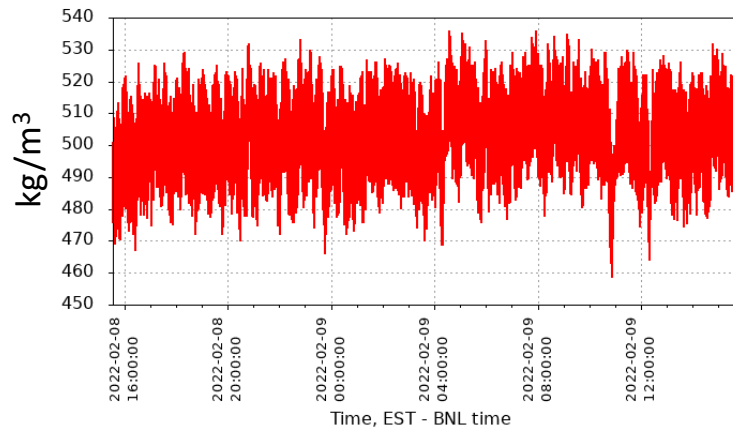
n-Pentane/CO2 Ratio

sTGC:Gas:GPA:ratio, %



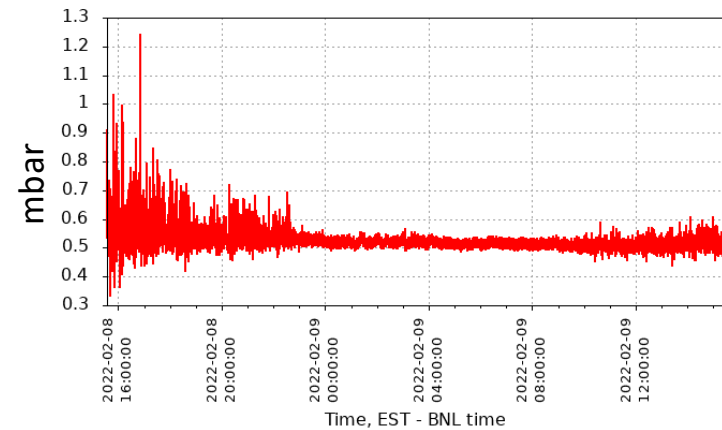
Chamber input pressure

sTGC:ADAM:PT-6:pressure, mbar



Measured density
of n-Pentane

sTGC:Gas:Penane:Density, kg/m3



Chamber vent pressure

sTGC:ADAM:PT-5:pressure, mbar

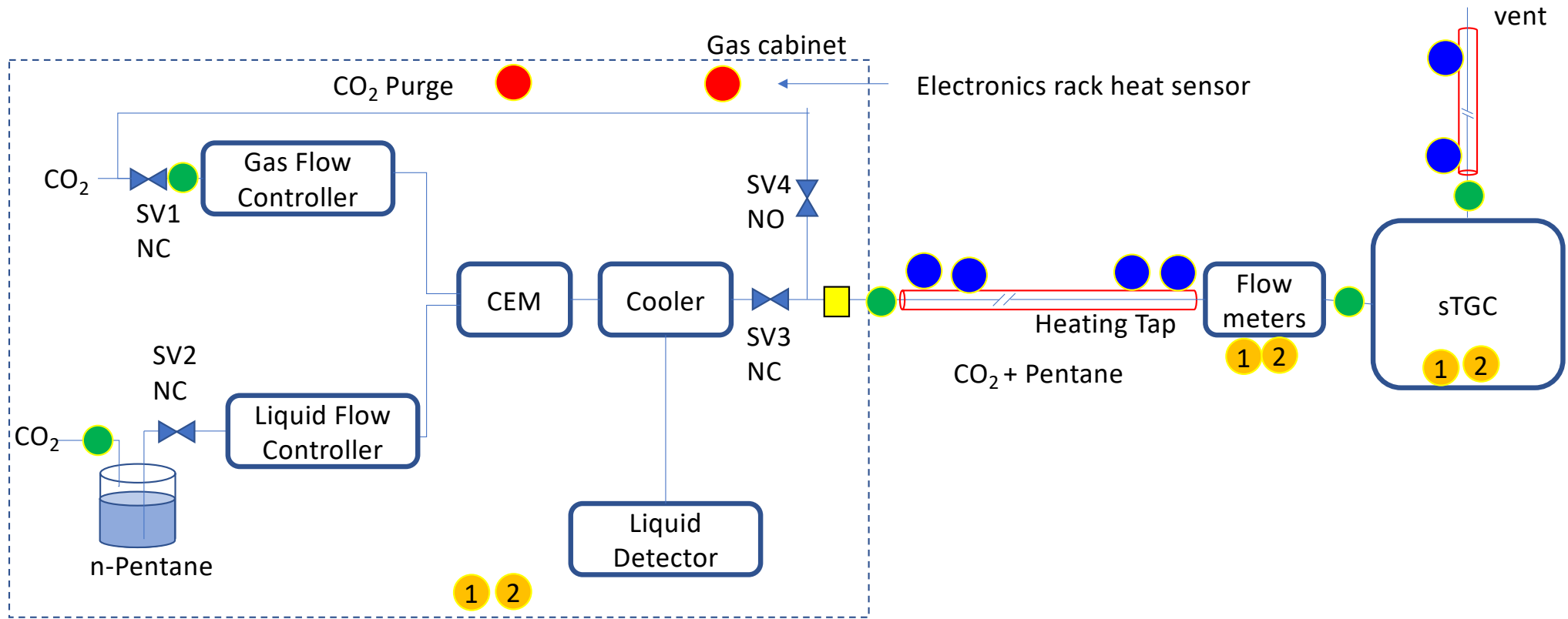
Building the Safety System

Safety System

		Status During Interlock					
		sTGC Purge	sTGC No Flow	sTGC LV Permissive	sTGC HV Permissive	UPS Power for Control Cabinet	Audible & Visible Alarm
1	Normal status	Mixing	Mixing	Enable	Enable	On	Off
Interlocks							
Fire/Heat Detection							
2	Heat in gas cabinet	X		X	X		X
3	Heat in electronic cabinet	X		X	X	X	
Pentane Gas Leak Detection							
4	15% of LEL in pentane sniffer 1 - Gas cabinet	X		X	X		X
5	15% of LEL in pentane sniffer 1 - Flow meters	X		X	X		X
6	15% of LEL in pentane sniffer 1- sTGC chambers	X		X	X		X
7	15% of LEL in pentane sniffer 2 - Gas cabinet	X		X	X		X
8	15% of LEL in pentane sniffer 2 - Flow meters	X		X	X		X
9	15% of LEL in pentane sniffer 2 - sTGC chambers	X		X	X		X
10	Pentane sniffer 1 malfunction w/5 min delay	X		X	X		X
11	Pentane sniffer 2 malfunction w/5 min delay	X		X	X		X
Gas mixing and Delivery							
12	Liquid pentane present after mixing	X		X	X		X
13	Supply line heat tap -LOW/HIGH	X		X	X		X
14	Vent line heat tap -LOW/HIGH	X		X	X		X
Pressure							
15	sTGC Supply over pressure (PT5)		X	X	X		X
STAR global interlock (SGIS)							
16	From SGIS	Appropriate action to be determined, not implemented for Run21					
17	To SGIS	Appropriate action to be determined, not implemented for Run21					

State Table

Safety Sensors



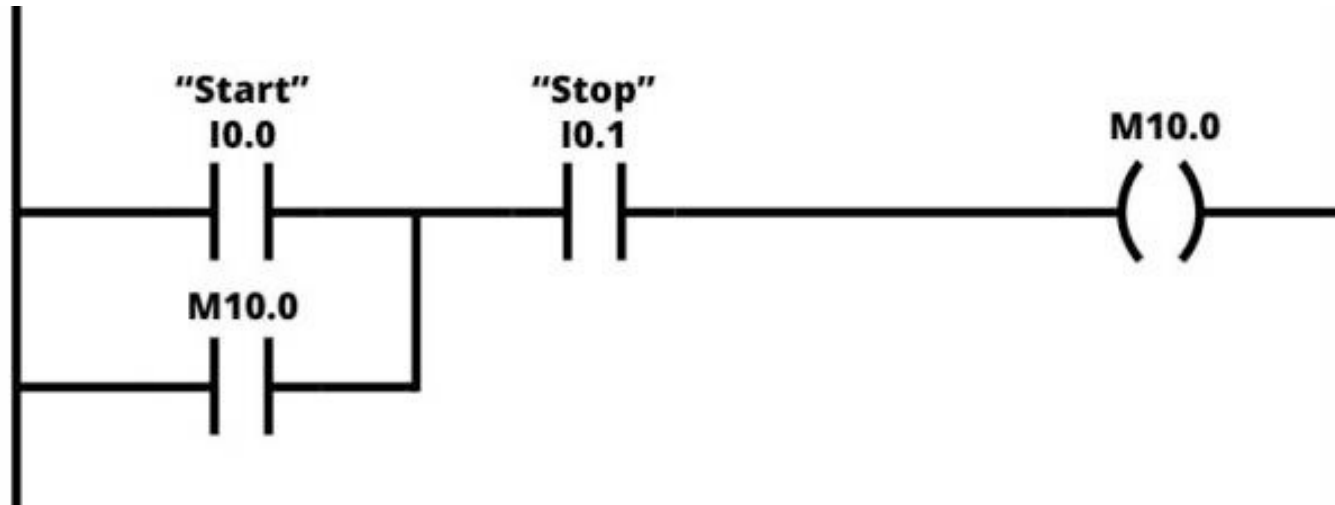
SV – Solenoid valves
NC – Normally closed
NO – Normally open

Yellow circle: Pentane Sniffer,
1 & 2 are independent monitoring
Red circle: Heat sensor

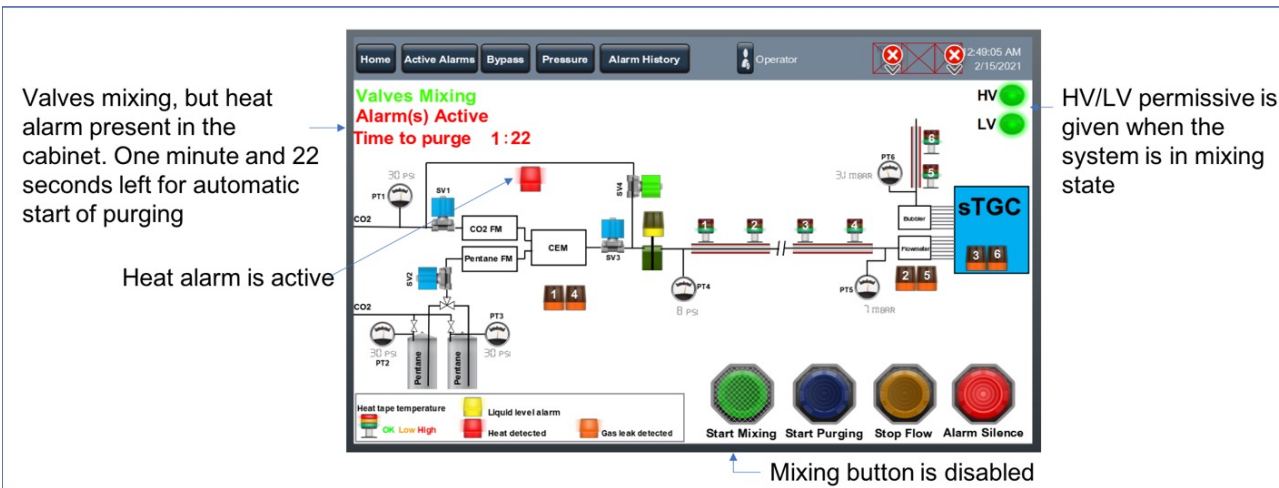
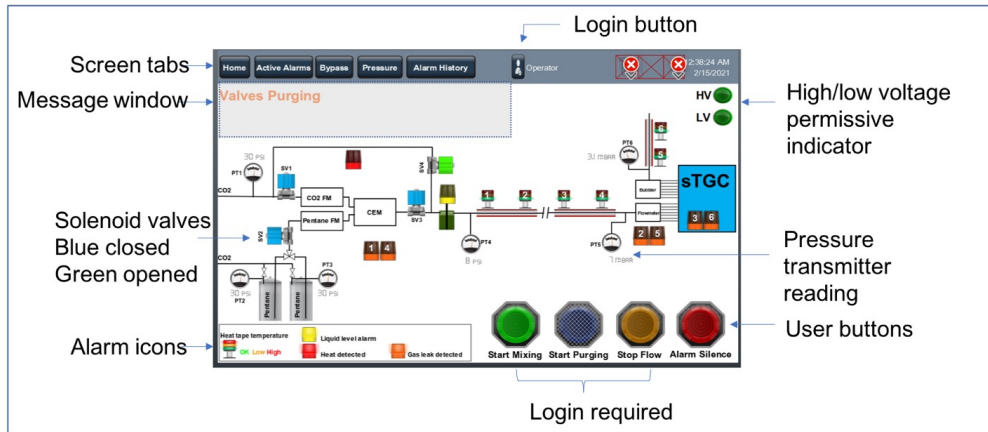
Blue circle: Thermocouple
Green circle: Pressure transmitter

Yellow square: Liquid detector

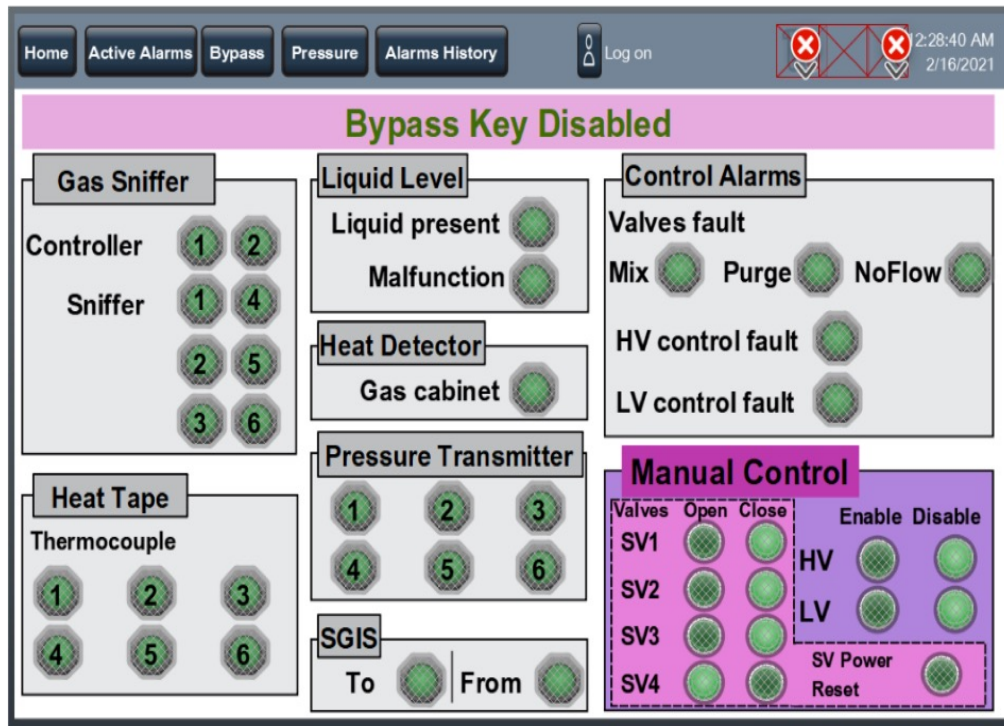
PLC – Ladder Diagram



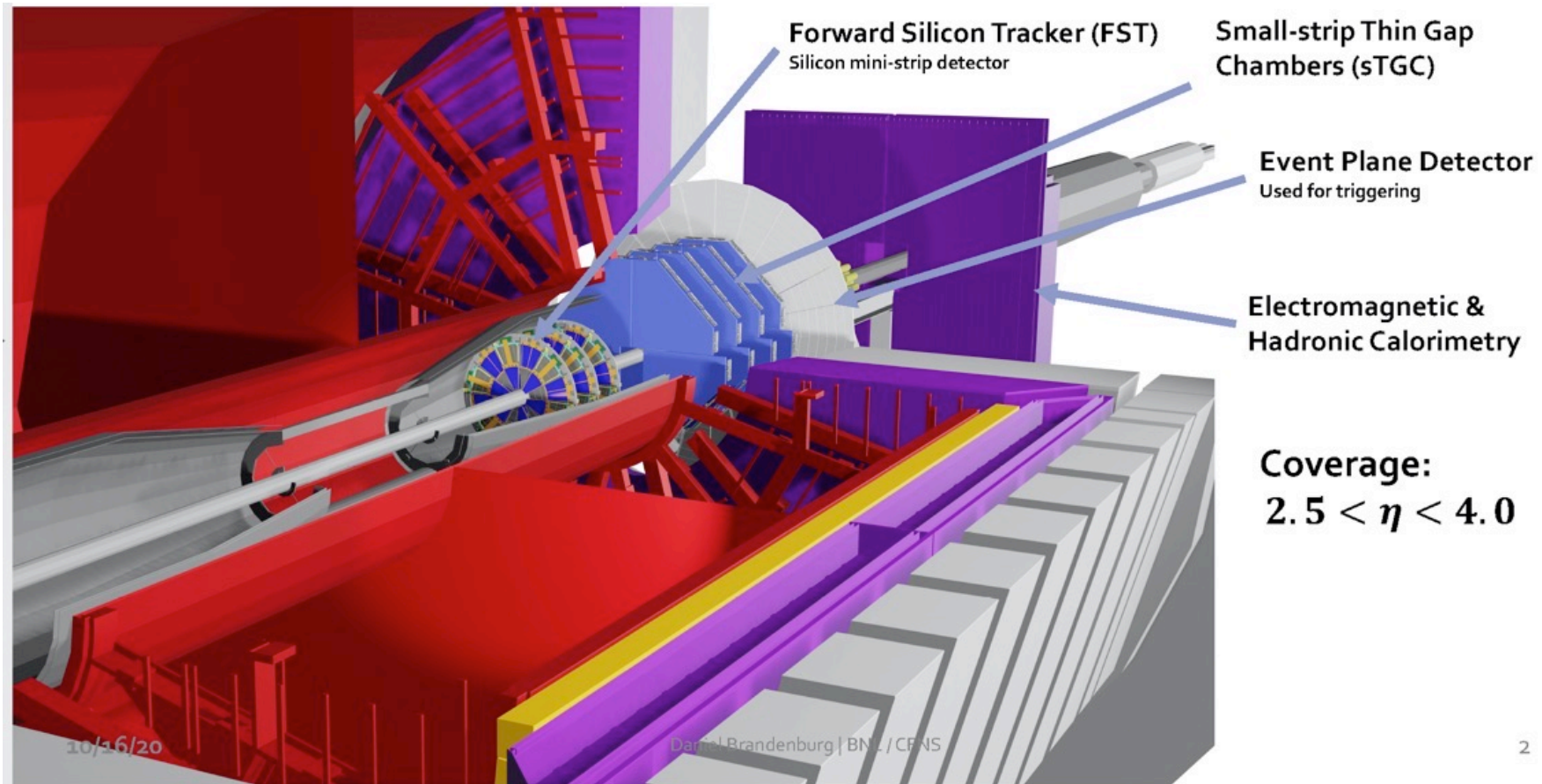
PLC - Controls

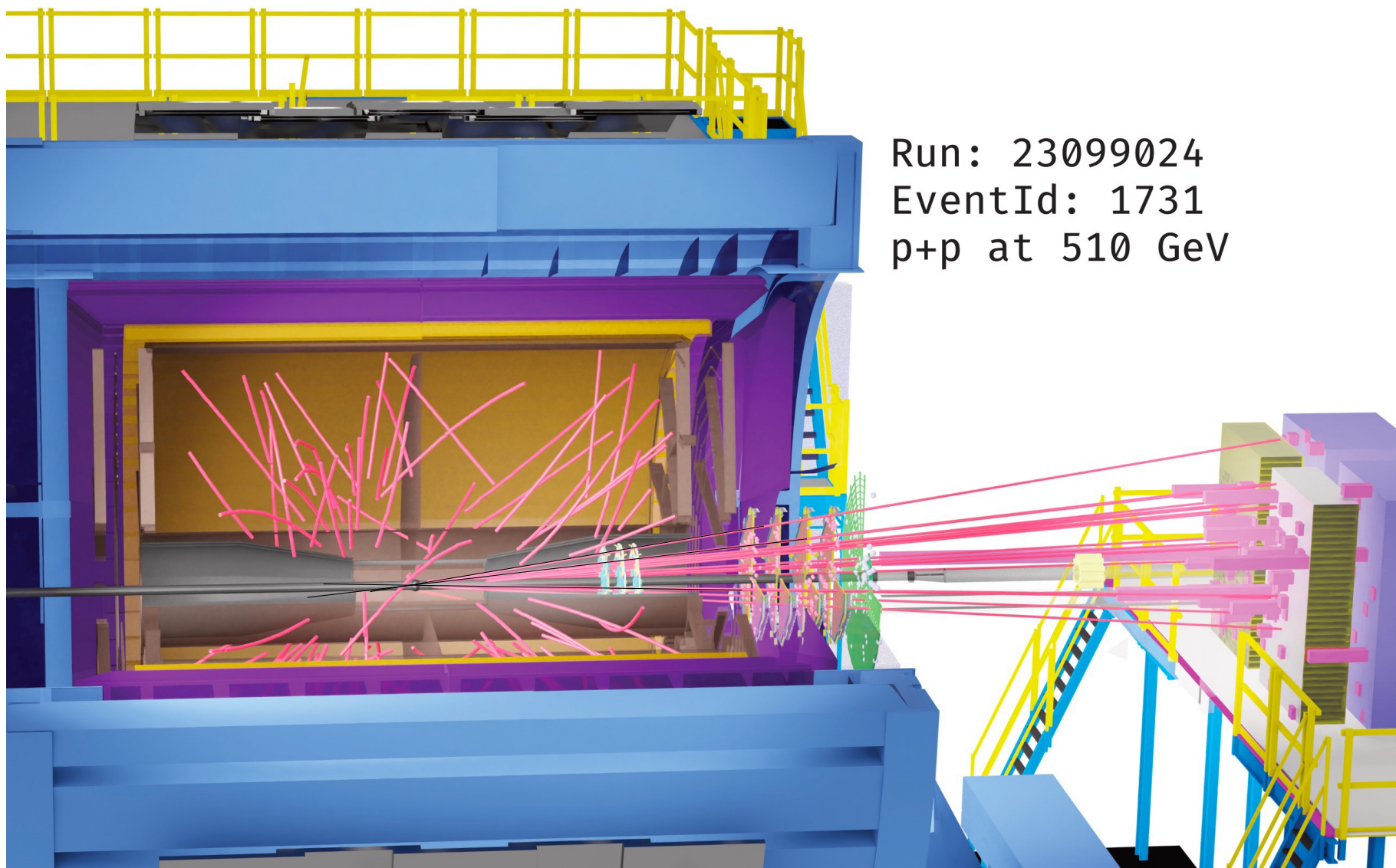


PLC - Controls



STAR Forward Upgrade





Homework

- Sketch a diagram for 10 mbar overpressure protector?
- A point charge q located near infinite grounded conducting plate, what are the
 - $E(r)$
 - $V(r)$