# Phase Evolution and Interfaces in Electrode Materials for Energy Storage



Dong Su(苏东) Center for Functional Nanomaterials, Brookhaven National Laboratory and Department of Materials Science and Engineering, Stony Brook University

Email: dsu@bnl.gov



### Energy Storage: Fuel Cell vs Batteries



# Outline

• Introduction:

Advanced Transmission Electron Microscopy

Structure-Property relation of Electrode Materials

 (i) Fe<sub>3</sub>O<sub>4</sub> nanoparticle and thin film for LIB

(ii) PtPb-Pt nanoplate catalyst for fuel cell

Outlook

# TEM: Transmission Electron Microscopy(e)











## STEM : Scanning Transmission Electron Microscopy





Li, Su, Sun et al. Nano Letters, 2015

# Electron energy-loss spectroscopy (EELS)



*Kim, Su, Wang, et al. ACS Nano, 2015* 5

# **Transmission Electron Microscopes at CFN**



JEOL 1400 JEOL 2100F FEI-Talos 200 Hitachi HD2700C Titan 80-300 - ETEM Capability:

Versatile analytical & in-situ TEM Analytical STEM STEM-EELS

Soft & biological materials

Analytical instrument: 3D STEM and STEM-EDX Environmental & in-situ TEM

# Approach: Combine Ex-situ and In-situ TEM



#### Correlating TEM results with other measurements

Huang et al. Science 330,1515 (2010) Lin *et al*. Nature Comm. 5, 3358 (2014). He, Su *et al*. Nano Lett. 15, 1437 (2015).

### Example : Reaction interface of Sodiation of NiO



He, et al. and Su, Nano Lett. 15, 5755 (2015).

### **Example:** In-situ TEM of Sodiation vs Lithiation

#### Sodiation

### Lithiation



### $NiO+Li^+/Na^+ + e^- \rightarrow Li_2O/Na_2O+Ni$

Same reaction different reaction process!

He, et al. and Su, Nano Lett. 15, 5755 (2015).

# Outline

- Introduction: Advanced Transmission Electron Microscopy
- Structure-Property relation of Electrode Materials
   (i) Fe<sub>3</sub>O<sub>4</sub> nanoparticle and thin film for LIB

(ii) PtPb-Pt nanoplate catalyst for ORR

Outlook

# Lithiation of Inverse Spinel Fe<sub>3</sub>O<sub>4</sub>



### 8 Li<sup>+</sup>+ 8 e<sup>-</sup>+Fe<sub>3</sub>O<sub>4</sub> $\rightarrow$ 4Li<sub>2</sub>O+ 3Fe

Capacity: 926 mAh/g

Samples from Chris Murray at Penn

# Lithiation of Fe<sub>3</sub>O<sub>4</sub>: Intermediate Phase



He, et al. Murray and Su, Nature Comm. 7, 11441 (2016)

# In-situ Electron Diffraction



## Strain Sensitive STEM Imaging: BF/LAADF



### Strain-sensitive: Bright-Field mode or Low-angle ADF mode

## **Comparison Between HAADF and LAADF**



Li, et al. and Su, ACS Nano, 10, 9577(2016)

# In-situ Bright-Field STEM



Fe<sub>3</sub>O<sub>4</sub>

LiFe<sub>3</sub>O<sub>4</sub>

Li<sub>2</sub>O +Fe

# HR TEM of Ex-situ/In-situ Samples



## In-situ BF-STEM: Reaction Inhomogeneity



• The lithiation of nanomaterial is highly inhomogeneous

He, et al. Murray and Su, Nature Comm. 7, 11441 (2016)

# Fe<sub>3</sub>O<sub>4</sub> Epitaxial Thin Film



Under compressive strain from substrate, partially relaxed by interfacial defects Samples from Ying-Hao Chu's group

Hwang, et al. and Su, Angewandte Chemie, (2017)

# In-situ STEM on Lithiation of Fe<sub>3</sub>O<sub>4</sub> Film



- Formation of cracks at upper film
- Non-conversion area close to Fe<sub>3</sub>O<sub>4</sub>/SrTiO<sub>3</sub> interface

## **Phase Identification**



• The formation of rock-salt phase close to interface

# Phase Field Simulations: Formula

### Lithium diffusion inside $Fe_3O_4$ : Cahn-Hilliard equation

$$\frac{\partial c}{\partial t} = \nabla M c \nabla (\Delta \mu)$$

Strain field in  $Fe_3O_4$  thin film described by van Der Merwe's t heory:

$$\varepsilon_{ii}^{(in)} = \varepsilon_0^{(in)} \exp(-ax)$$

Strain coupling:

$$f_{el_c} = \frac{1}{2} K \varepsilon_{ii}^{(in)} \varepsilon_{ii}^{(0)}$$

J. Newman, *Electrochemical Systems*, Prentice Hall, **1991**.
J. H. Van der Merwe, *Proc. Phys. Soc.* 63, 616–637,(1950)
M. Tang, *et al.* and Y.-M. Chiang, *Chem. Mater.* 21, 1557–1571, (2009)

# Phase Field Simulations: Surface vs. Bulk



Hwang, et al. and Su, Angewandte Chemie, DOI: 10.1002/ange.201703168, (2017)

# Outline

- Introduction: Advanced Transmission Electron Microscopy
- Structure-Property relation of Electrode Materials
   (i) Fe<sub>3</sub>O<sub>4</sub> nanoparticle and thin film for LIB

(ii)PtPb-Pt nanoplate catalyst for fuel cell

Outlook

# PEMFC: Proton-exchange membrane fuel cell



ORR:  $O_2 + 4H^+ \rightarrow 2H_2O$ 

# Fuel Cell: Target vs Reality

#### □ Fuel Cell Device



#### State of the art of Pt nanoparticle

- □ Slow kinetics for ORR
  - Mass activity: ~0.11 A/mg
  - Specific activity: ~0.2 mA/cm<sup>2</sup>

#### Durability



Ref: Wang et al., Nano Lett. 11, 919(2011)

#### **2020 DOE Technical Targets**

- Mass activity@ 0.9 V: ~0.44 A/mg
- Specific activity @ 0.9 V : ~0.72 mA/cm<sup>2</sup>
- Electrochemical area loss: < 40%</p>
- Catalyst support loss: < 30%</p>
- PGM Total loading: 0.2 mg/cm<sup>2</sup> electrode
- Durability w/cycling (80 °C): 5000 hrs

## Pt-based Multimetallic Catalysis

Alloying/Core shell

#### Shape control



Ref.: Markovic *et al.*, *Nat. Mater.* 7, 241,(2007) Markovic *et al.*, *Science*, 315, 493,(2007)

# **Optimization of Nanostructures**

#### A: PtNi nanoframes



Yang and Markovic 's groups, Science 2014, 343, 1399

#### **C: Ordered structure of Pt<sub>3</sub>Co**



Y. N. Xia's group, *Science* 2015, 349, 412



D: Stacking sequence of Pt(fcc)-Ru(hcp)



Abruna's group, Nature Materials, 2013,12,81

With Jia Wang and R. Adzic, Nature Comm., 2013,4,2466

# PtPb-Pt Core-Shell Plate for ORR



• Excellent activity and stability! But why?

Synthesis and Electrochemical results by Huang's and Guo's groups

# PtPb-Pt core-shell Nanoplate : Hexagonal@Fcc



- Core: PtPb/HCP vs Shell: Pt/FCC
- Interfacial coherence:
- Surface Pt and edge Pt
- Corner dislocations

# Strain Analysis from Diffraction Patterns

#### Experimental results(FFT)



#### Simulations



## PtPb-Pt Core-Shell: Biaxial Strain



Guo, Su, Huang et al. Science 354,1410(2016)

# DFT Calculation: Binding Energy and Strain



DFT calculation by Xu zhang and Gang Lu

# Outline

- Introduction: Advanced Transmission Electron Microscopy
- Structure-Property relation of Electrode Materials
   (i) Fe<sub>3</sub>O<sub>4</sub> nanoparticle and thin film for LIB

(ii)PtPb-Pt nanoplate catalyst for fuel cell

Outlook

# Outlook on In-situ/Operando TEM



- To reveal real-time and spatially-resolved m orphological, structural, chemical, and elect ronic state evolutions during physical and c hemical processes.
- To probe direct material response to multip le stimulus applied to the nanoscale system.
- To combine complimentary methodologies simultaneously and at various relevant leng th scales, enabling information acquisition i n extra dimensions.



Ref. H. Zheng et al., MRS Bulletin, 2015

# Acknowledgement



#### • TEM work:

Dr. Kai He(now AP at Clemson), Dr. Eric Stach(Now at Penn), Dr. Jing Li(BNL/SBU), Dr. Sooyeon Hwang(BNL), Mr. K. Kisslinger(BNL), Dr. Huolin Xin(BNL), Dr. Yimei Zhu (BNL),

#### • Materials and tests:

Dr. Yiqian Yu(IOP), Dr. Xiao-qiang Yang(BNL), Prof. Gerb Ceder(Berkeley), Prof. Xin Li(Harvard), Prof. Sen Zhang (Virginia), Prof. Chris Murray (Penn), Dr . Feng Lin and Dr. Marca Doeff (LBNL), Prof. Ryan Richards(UC Mines) Dr. Ratko Adzic(BNL), Prof. Shouheng Sun(Brown), Prof. Shaojun Guo(Peking), Prof. Minhua Shao(HKUST), Prof. Xiaowei Teng(New Hampshire), Prof. Xiaoqing Huang(Soochow), Prof. Gang Wu(Buffalo)

#### • Theoretical calculations:

Prof. Yifei Mo(Maryland) Dr. Qingping Meng (BNL), Dr. Xu Zhang and Prof. Gang Lu(California State), Prof. Ju Li(MIT), Prof. Kejie Zhao(Purdue)

# Thank you for your attention, 谢谢

